

Gate Dielectric Chemical Structure—Organic Field-Effect Transistor Performance Correlations for Electron, Hole, and Ambipolar Organic Semiconductors

Myung-Han Yoon, Choongik Kim, Antonio Facchetti,* and Tobin J. Marks*

Contribution from the Department of Chemistry and the Materials Research Center,
Northwestern University 2145 Sheridan Road, Evanston, Illinois, 60208

Received May 11, 2006; E-mail: a-facchetti@northwestern.edu; t-marks@northwestern.edu

Abstract: This study describes a general approach for probing semiconductor–dielectric interfacial chemistry effects on organic field-effect transistor performance parameters using bilayer gate dielectrics. Organic semiconductors exhibiting p/n-type or ambipolar majority charge transport are grown on six different bilayer dielectric structures consisting of various spin-coated polymers/HMDS on 300 nm SiO₂/p⁺-Si, and are characterized by AFM, SEM, and WAXRD, followed by transistor electrical characterization. In the case of air-sensitive (generally high LUMO energy) n-type semiconductors, dielectric surface modifications induce large variations in the corresponding OTFT performance parameters although the film morphologies and microstructures remain similar. In marked contrast, the device performance of air-stable n-type and p-type semiconductors is not significantly affected by the same dielectric surface modifications. Among the bilayer dielectric structures examined, nonpolar polystyrene coatings on SiO₂ having minimal gate leakage and surface roughness significantly enhance the mobilities of overlying air-sensitive n-type semiconductors to as high as ~ 2 cm²/(V s) for α,ω -diperfluorohexylcarbonylquaterthiophene polystyrene/SiO₂. Electron trapping due to silanol and carbonyl functionalities at the semiconductor–dielectric interface is identified as the principal origin of the mobility sensitivity to the various surface chemistries in the case of n-type semiconductors having high LUMO energies. Thiophene-based n-type semiconductors exhibiting similar film morphologies and microstructures on various bilayer gate dielectrics therefore provide an incisive means to probe TFT performance parameters versus semiconductor–dielectric interface relationships.

Introduction

Field-effect-active organic semiconductors (OSCs)^{1,2} are of great interest for use in low-cost/disposable electronic products such as smart cards and radio frequency identification (RFID)

tags,³ as well as in flexible display driver circuits,⁴ nonvolatile memories,⁵ and sensors.⁶ Indeed, amorphous and polycrystalline films of several OSCs exhibit hole or electron carrier mobilities

- (1) For recent reviews on this topic, see: (a) Chabinyk, M.; Loo, Y.-L. *J. Macromol. Sci. Polym. Rev.* **2006**, *46*, 1–5. (b) Dodabalapur, A. *Nature* **2005**, *434*, 151. (c) Sirringhaus, H. *Adv. Mater.* **2005**, *17*, 2411. (d) Newman, C. R.; Frisbie, C. D.; da Silva Filho, D. A.; Bredas, J.-L.; Ewbank, P. C.; Mann, K. R. *Chem. Mater.* **2004**, *16*, 4436. (e) Horowitz, G. *J. Mater. Res.* **2004**, *19*, 1946. (f) Dimitrakopoulos, C. D.; Malenfant, P. R. L. *Adv. Mater.* **2002**, *14*, 99. (g) Rogers, J. A.; Bao, Z.; Katz, H. E.; Dodabalapur, A. *Thin-Film Transistors*; Kagan, C., Andry R. P., Eds; Marcel Dekker, Inc.: New York, USA, 2003; pp 377–425.
- (2) (a) Kim, D. H.; Han, J. T.; Park, Y. D.; Jang, Y.; Cho, J. H.; Hwang, M.; Cho, K. *Adv. Mater.* **2006**, *18*, 719. (b) Tulevski, G. S.; Miao, Q.; Afzali, A.; Graham, T. O.; Kagan, C. R.; Nuckolls, C. *J. Am. Chem. Soc.* **2006**, *128*, 1788. (c) Ponomarenko, S. A.; Kirchmeyer, S.; Elschner, A.; Alpatova, N. M.; Halik, M.; Klauk, H.; Zschieschang, U.; Schmid, G. *Chem. Mater.* **2006**, *18*, 579. (d) Ando, S.; Nishida, J.; Tada, H.; Inoue, Y.; Tokito, S.; Yamashita, Y. *J. Am. Chem. Soc.* **2005**, *127*, 5336. (e) Ofuji, M.; Lovinger, A. J.; Kloc, C.; Siegrist, T.; Maliakal, A. J.; Katz, H. E. *Chem. Mater.* **2005**, *17*, 5748. (f) Wu, Y.; Ong, B. S. *J. Am. Chem. Soc.* **2005**, *127*, 3266. (g) Murphy, A. R.; Liu, J.; Luscombe, C.; Kavulak, D.; Frechet, J. M. J.; Kline, R. J.; McGehee, M. D. *Chem. Mater.* **2005**, *17*, 4892. (h) Locklin, J.; Li, D.; Mannsfeld, S. C. B.; Borkent, E.-J.; Meng, H. *Adv. Mater.* **2005**, *17*, 3366. (i) Payne, M. M.; Parkin, S. R.; Anthony, J. E. *J. Am. Chem. Soc.* **2005**, *127*, 8028. (j) Swartz, C. R.; Parkin, S. R.; Bullock, J. E.; Anthony, J. E.; Mayer, A. C.; Malliaras, G. G. *Org. Lett.* **2005**, *7*, 3163. (k) Payne, M. M.; Parkin, S. R.; Anthony, J. E.; Kuo, C.-C.; Jackson, T. N. *J. Am. Chem. Soc.* **2005**, *127*, 4986. (l) Sakamoto, Y.; Suzuki, T.; Kobayashi, M.; Gao, Y.; Fukui, Y.; Inoue, Y.; Sato, F.; Tokito, S. *J. Am. Chem. Soc.* **2004**, *126*, 8138. (m) Kunugi, Y.; Takimiya, K.; Toyoshima, Y.; Yamashita, K.; Aso, Y.; Otsubo, T. *J. Mater. Chem.* **2004**, *14*, 1367.
- (3) (a) Yoo, B.; Jung, T. O.; Basu, D.; Dodabalapur, A.; Jones, B. A.; Facchetti, A.; Wasielewski, M. R.; Marks, T. J. *Appl. Phys. Lett.* **2006**, *88*, 082104. (b) Subramanian, V.; Chang, P. C.; Lee, J. B.; Moles, S. E.; Volkman, S. K. *IEEE Trans. Comp. Pack. Tech.* **2005**, *28*, 742. (c) Baude, P. F.; Ender, D. A.; Haase, M. A.; Kelley, T. W.; Muyses, D. V.; Theiss, S. D. *Appl. Phys. Lett.* **2003**, *82*, 3964. (d) Drury, C. J.; Mutsaers, C. M. J.; Hart, C. M.; Matters, M.; de Leeuw, D. M. *Appl. Phys. Lett.* **1998**, *73*, 108. (e) Gamota, D. R.; Brazis, P.; Kalyanasundaram, X.; Zhang, J. *Printed Organic and Molecular Electronics*; Kluwer Academic Publishers: New York, USA, 2004.
- (4) (a) Street, R. A.; Wong, W. S.; Ready, S. E.; Chabinyk, M. L.; Arias, A. C.; Limb, S.; Salleo, A.; Lujan, R. *Mater. Today (Oxford, U. K.)* **2006**, *9*, 32. (b) Zhou, L.; Wang, A.; Wu, S.-C.; Sun, J.; Park, S.; Jackson, T. N. *Appl. Phys. Lett.* **2006**, *88*, 083502/1. (c) Zyung, T.; Kim, S. H.; Lim, S. C.; Lee, J. H.; Chu, H. Y.; Lee, J.-I.; Oh, J. Y. *J. Korean Phys. Soc.* **2006**, *48*, S111. (d) Oh, S. Y.; Lee, J.-Y. *Mol. Cryst. Liq. Cryst.* **2006**, *444*, 211. (e) Chabinyk, M. L.; Salleo, A. *Chem. Mater.* **2004**, *16*, 4509. (f) Sheraw, C. D.; Zhou, L.; Huang, J. R.; Gundlach, D. J.; Jackson, T. N.; Kane, M. G.; Hill, I. G.; Hammond, M. S.; Campi, J.; Greening, B. K.; Francl, West, J. J. *Appl. Phys. Lett.* **2002**, *80*, 1088. (g) Mach, P.; Rodriguez, S. J.; Nortrup, R.; Wiltzius, P.; Rogers, J. A. *Appl. Phys. Lett.* **2001**, *78*, 3592. (h) Klauk, H.; Gundlach, D. J.; Nichols, J. A.; Jackson, T. N. *IEEE Trans. Electron Devices* **1999**, *46*, 1258.
- (5) Naber, R. C. G.; de Boer, B.; Blom, P. W. M.; de Leeuw, D. M. *Appl. Phys. Lett.* **2005**, *87*, 203509/1.
- (6) (a) Someya, T.; Sekitani, T.; Iba, S.; Kato, Y.; Sakurai, T.; Kawaguchi, H. *Mol. Cryst. Liq. Cryst.* **2006**, *444*, 13. (b) Wang, L.; Fine, D.; Sharma, D.; Torsi, L.; Dodabalapur, A. *Anal. Bioanal. Chem.* **2006**, *384*, 310. (c) Torsi, L. *Anal. Bioanal. Chem.* **2006**, *384*, 309. (d) Zhu, Z.-T.; Mason, J. T.; Dieckmann, R.; Malliaras, G. G. *Appl. Phys. Lett.* **2002**, *81*, 4643. (e) Someya, T.; Katz, H. E.; Gelperin, A.; Lovinger, A. J.; Dodabalapur, A. *Appl. Phys. Lett.* **2002**, *81*, 3079.

comparable to or surpassing⁷ those of the common inorganic semiconductor for the aforementioned applications: amorphous hydrogenated silicon (a-Si:H).⁸ As an example, this material is currently used in fabricating thin film transistors (TFTs) for LC/LED displays and exhibits an electron carrier mobility of $\sim 1.0 \text{ cm}^2/(\text{V s})$ and $I_{\text{on}}/I_{\text{off}}$ ratio $> 10^6$. For OSCs, such levels of TFT performance have been achieved by continuous innovation in molecular design⁹ as well as by careful control of vapor-/solution-phase film growth conditions via optimization of substrate temperature, solvent, deposition/solvent evaporation rate, material purity, etc.¹⁰ It is generally accepted that all of the aforementioned factors strongly influence semiconductor film microstructure, which in turn heavily influences charge transport.¹¹ Other critical factors affecting overall semiconductor/device performance are chemical composition, surface functionalization, and surface morphology (roughness) of the gate dielectric.^{12,13}

Pioneering studies by Horowitz and Garnier¹⁴ investigated the consequences for oligothiophene-based TFT response of employing various polymeric insulators, instead of SiO_2 , as the gate dielectric layer. This work was followed by more recent investigations.^{15,16} For example, Sirringhaus et al. showed that implementation of appropriate insulators facilitates electron

transport for typical organic p-type semiconductors. More recently, it was also reported that certain insulators enable ambipolar transport for pentacene- and rubrene-based OTFTs (organic TFTs).¹⁷ Regarding OTFT dielectric–semiconductor interfacial effects, most studies have employed SiO_2 as the dielectric and pentacene as the semiconductor.^{18–28} It is generally accepted that gate dielectric surface roughness is an important parameter affecting OTFT electrical performance,¹⁸ and it was shown that rougher gate dielectric surfaces result in smaller pentacene grains and lower OTFT carrier mobilities.¹⁹ Note that several groups have explored correlations between pentacene grain size (tuned by varying film deposition rate, substrate temperature) and charge mobility, with most reporting increased mobility with increased pentacene grain size, although many aspects remain controversial.^{20,21} Very recently, the

- (7) (a) Lee, K. J.; Motala, M. J.; Meitl, M. A.; Childs, W. R.; Menard, E.; Shim, A. K.; Rogers, J. A.; Nuzzo, R. G. *Adv. Mat.* **2005**, *17*, 2332. (b) Sele, C. W.; von Werne, T.; Friend, R. H.; Sirringhaus, H. *Adv. Mater.* **2005**, *17*, 997. (c) Chabiny, M. L.; Salleo, A.; Wu, Y.; Liu, P.; Ong, B. S.; Heeney, M.; McCulloch, I. *J. Am. Chem. Soc.* **2004**, *126*, 13928. (d) Ling, M. M.; Bao, Z. *Chem. Mater.* **2004**, *16*, 4824. (e) Lefenfeld, M.; Blanchet, G.; Rogers, J. A. *Adv. Mater.* **2003**, *15*, 1188.
- (8) (a) Kuo, Y. *Thin Film Transistors*; Kluwer Academic Publishers: Norwell, MA, USA, 2004; pp 183–202. (b) Sazonov, A.; Striakhilev, D.; Lee, C.-H.; Nathan, A. *Proc. IEEE* **2005**, *93*, 1420. (c) Hamakawa, Y. *Ren. Energy* **1996**, *8*, 10.
- (9) (a) Videlot-Ackermann, C.; Ackermann, J.; Brisset, H.; Kawamura, K.; Yoshimoto, N.; Raynal, P.; El Kassmi, A.; Fages, F. *J. Am. Chem. Soc.* **2005**, *127*, 16343. (b) Wex, B.; Kaafarani, B. R.; Schroeder, R.; Majewski, L. A.; Burckel, P.; Grell, M.; Neckers, D. *J. Mater. Chem.* **2006**, *16*, 1121. (c) Crouch, D. J.; Skabara, P. J.; Lohr, J. E.; McDouall, J. J. W.; Heeney, M.; McCulloch, I.; Sparrow, D.; Shkunov, M.; Coles, S. J.; Horton, P. N.; Hursthouse, M. B. *Chem. Mater.* **2006**, *18*, 6567. (d) Zen, A.; Bilge, A.; Galbrecht, F.; Alle, R.; Meerholz, K.; Grenzer, J.; Neher, D.; Scherff, U.; Farrell, T. *J. Am. Chem. Soc.* **2006**, *128*, 3914. (e) Sun, Y.; Ma, Y.; Liu, Y.; Lin, Y.; Wang, Z.; Wang, Y.; Di, C.; Xiao, K.; Chen, X.; Qiu, W.; Zhang, B.; Yu, G.; Hu, W.; Zhu, D. *Adv. Funct. Mater.* **2006**, *16*, 426. (f) Takimiya, K.; Kunugi, Y.; Konda, Y.; Ebata, H.; Toyoshima, Y.; Otsubo, T. *J. Am. Chem. Soc.* **2006**, *128*, 3044. (g) Zhang, H.; Wang, Y.; Shao, K.; Liu, Y.; Chen, S.; Qiu, W.; Sun, X.; Qi, T.; Ma, Y.; Yu, G.; Su, Z.; Zhu, D. *Chem. Commun.* **2006**, 755.
- (10) (a) Locklin, J.; Roberts, M.; Mannsfeld, S.; Bao, Z. *J. Macromol. Sci. Polym. Rev.* **2006**, *46*, 79. (b) Jurchescu, O. D.; Baas, J.; Palstra, T. T. M. *Appl. Phys. Lett.* **2004**, *84*, 3061. (c) Shtein, M.; Mapel, J.; Benziger, J. B.; Forrest, S. R. *Appl. Phys. Lett.* **2002**, *81*, 268. (d) Jentzsch, T.; Juepner, H. J.; Brzezinka, K. W.; Lau, A. *Thin Solid Films* **1998**, *315*, 273. (e) Dimitrakopoulos, C. D.; Brown, A. R.; Pomp, A. *J. Appl. Phys.* **1996**, *80*, 2501. (f) Laquindanum, J. G.; Katz, H. E.; Lovinger, A. J.; Dodabalapur, A. *Chem. Mater.* **1996**, *8*, 2542.
- (11) (a) Naraso; Nishida, J.-i.; Ando, S.; Yamaguchi, J.; Itaka, K.; Koinuma, H.; Tada, H.; Tokito, S.; Yamashita, Y. *J. Am. Chem. Soc.* **2005**, *127*, 10142. (b) DeLongchamp, D. M.; Sambasivan, S.; Fischer, D. A.; Lin, E. K.; Chang, P.; Murphy, A. R.; Frechet, J. M. J.; Subramanian, V. *Adv. Mater.* **2005**, *17*, 2340. (c) Abthagir, P. S.; Ha, Y.-G.; You, E.-A.; Jeong, S.-H.; Seo, H.-S.; Choi, J.-H. *J. Phys. Chem. B* **2005**, *109*, 23918. (d) Porzio, W.; Destri, S.; Giovannella, U.; Pasini, M.; Motta, T.; Natali, D.; Sampietro, M.; Campione, M. *Thin Solid Films* **2005**, *492*, 212. (e) Singh, Th. B.; Gunes, S.; Marjanovic, N.; Sariciftci, N. S.; Menon, R. *J. Appl. Phys.* **2005**, *97*, 114508/1. (f) Facchetti, A.; Letizia, J.; Yoon, M.-H.; Musherush, M.; Katz, H. E.; Marks, T. J. *Chem. Mater.* **2004**, *16*, 4715. (g) Meng, H.; Zheng, J.; Lovinger, A. J.; Wang, B.-C.; Van Patten, P. G.; Bao, Z. *Chem. Mater.* **2003**, *15*, 1778. (h) Videlot, C.; Ackermann, J.; Blanchard, P.; Raimundo, J.-M.; Frere, P.; Allain, M.; de Bettignies, R.; Levillain, E.; Roncali, J. *Adv. Mater.* **2003**, *15*, 306. (i) Videlot, C.; Ackermann, J.; El Kassmi, A.; Raynal, P. *Thin Solid Films* **2002**, *403*, 380. (j) Chen, X. L.; Lovinger, A. J.; Bao, Z.; Sapjeta, J. *Chem. Mater.* **2001**, *13*, 1341. (k) Horowitz, G.; Hajlaoui, M. E. *Synth. Met.* **2001**, *122*, 185. (l) Katz, H. E. *J. Mater. Chem.* **1997**, *7*, 369.
- (12) (a) Facchetti, A.; Yoon, M.-H.; Marks, T. J. *Adv. Mater.* **2005**, *17*, 1705. (b) Veres, J.; Ogier, S.; Lloyd, G.; de Leeuw, D. *Chem. Mater.* **2004**, *16*, 4543.
- (13) (a) Jung, T.; Dodabalapur, A.; Wenz, R.; Mohapatra, S. *Appl. Phys. Lett.* **2005**, *87*, 182109/1. (b) Backlund, T. G.; Osterbacka, R.; Stubb, H.; Bobacka, J.; Ivaska, A. *J. Appl. Phys.* **2005**, *98*, 074504/1. (c) Panzer, M. J.; Frisbie, C. D. *J. Am. Chem. Soc.* **2005**, *127*, 6960. (d) Maliakal, A.; Katz, H. E.; Cotts, P.; Subramoney, M.; Mirau, S. P. *J. Am. Chem. Soc.* **2005**, *127*, 14655. (e) Schroeder, R.; Majewski, L. A.; Grell, M. *Adv. Mater.* **2005**, *17*, 1535. (f) Wang, G.; Moses, D.; Heeger, A. J.; Zhang, H.-M.; Narasimhan, M.; Demaray, R. E. *J. Appl. Phys.* **2004**, *95*, 316. (g) Dimitrakopoulos, C. D.; Purushothaman, S.; Kymissis, J.; Callegari, A.; Shaw, J. M. *Science* **1999**, *283*, 822.
- (14) Peng, X.; Horowitz, G.; Fichou, D.; Garnier, F. *Appl. Phys. Lett.* **1990**, *57*, 2013.
- (15) (a) Jang, Y.; Kim, D. H.; Park, Y. D.; Cho, J. H.; Hwang, M.; Cho, K. *Appl. Phys. Lett.* **2006**, *88*, 072101. (b) Dinelli, F.; Moulin, J.-F.; Loi, M. A.; Da Como, E.; Massi, M.; Murgia, M.; Muccini, M.; Biscarini, F.; Wie, J.; Kingshot, P. *J. Phys. Chem. B* **2006**, *110*, 258. (c) Nunes, G., Jr.; Zane, S. G.; Meth, J. S. *J. Appl. Phys.* **2005**, *98*, 104503/1. (d) Knipp, D.; Kumar, P.; Voelkel, A. R.; Street, R. A. *Synth. Met.* **2005**, *155*, 485. (e) Naber, R. C.; Blom, P. W. M.; Gelinck, G. H.; Marsman, A. W.; de Leeuw, D. M. *Adv. Mater.* **2005**, *17*, 2692. (f) Stadlober, B.; Zirkl, M.; Beutl, M.; Leising, G.; Bauer-Gogonea, S.; Bauer, S. *Appl. Phys. Lett.* **2005**, *86*, 242902/1. (g) Klauk, H.; Halik, M.; Zschieschang, U.; Schmid, G.; Radlik, W.; Weber, W. *J. Appl. Phys.* **2002**, *92*, 5259.
- (16) Chua, L.-L.; Zausmell, J.; Chang, J.-F.; Ou, E. C.-W.; Ho, P. J.-H.; Sirringhaus, H.; Friend, R. H. *Nature* **2005**, *434*, 194.
- (17) (a) Singh, T. B.; Meghdadi, F.; Gunes, S.; Marjanovic, N.; Horowitz, G.; Lang, P.; Bauer, S.; Sariciftci, N. S. *Adv. Mater.* **2005**, *17*, 2315. (b) Takahashi, T.; Takenobu, T.; Takeya, J.; Iwasa, Y. *Appl. Phys. Lett.* **2006**, *88*, 033505/1.
- (18) (a) Knipp, D.; Street, R. A.; Volkel, A.; Ho, A. *J. Appl. Phys.* **2003**, *93*, 347. (b) Knipp, D.; Street, R. A.; Krusor, B. S.; Apte, R. B.; Ho, J. *Proc. SPIE-Int. Soc. Opt. Eng.* **2001**, *4466*, 8.
- (19) Stuedel, S.; De Vusser, S.; De Jonge, S.; Janssen, D.; Verlaak, S.; Genoe, P.; Heremans, J. *Appl. Phys. Lett.* **2004**, *85*, 4400.
- (20) (a) Ruiz, R.; Choudhary, D.; Nickel, B.; Toccoli, T.; Chang, K.-C.; Mayer, A. C.; Clancy, P.; Blakely, J. M.; Headrick, R. L.; Iannotta, S.; Malliaras, G. G. *Chem. Mater.* **2004**, *16*, 4497. (b) Knipp, D.; Murti, D. K.; Krusor, B.; Apte, R.; Jiang, L.; Lu, J. P.; Ong, B. S.; Street, R. A. *Mater. Res. Soc. Symp. Proc.* **2002**, *665*, 207. (c) Horowitz, G.; Hajlaoui, M. E. *Adv. Mater.* **2000**, *12*, 1046. (d) Klauk, H.; Jackson, T. N. *Solid State Technol.* **2000**, *43*, 63. (e) Lin, Y.-Y.; Gundlach, D. J.; Nelson, S. F.; Jackson, T. N. *IEEE Trans. Electron Devices* **1997**, *44*, 1325.
- (21) (a) Yang, S. Y.; Shin, K.; Park, C. E. *Adv. Funct. Mater.* **2005**, *15*, 1806. (b) Gundlach, D. J.; Kuo, C.-C.; Nelson, S. F.; Jackson, T. N. *IEEE 57th Device Research Conference Digest*; Santa Barbara, CA; IEEE Publishing Services: New York, USA 1999; pp 164–165.
- (22) Fritz, S. E.; Kelly, T. W.; Frisbie, C. D. *J. Phys. Chem. B* **2005**, *109*, 10574.
- (23) (a) Salleo, A.; Chabiny, M. L.; Yang, M. S.; Street, R. A. *Appl. Phys. Lett.* **2002**, *81*, 4383. (b) Lin, Y. Y.; Gundlach, D. J.; Nelson, S. F.; Jackson, T. N. *IEEE Electron Devices Lett.* **1997**, *18*, 606.
- (24) Yagi, I.; Tsukagoshi, K.; Aoyagi, W. *Appl. Phys. Lett.* **2005**, *86*, 103502-1.
- (25) (a) Kobayashi, S.; Nishikawa, T.; Takenobu, T.; Mori, S.; Shimoda, T.; Mitani, T.; Shimotani, H.; Yoshimoto, N.; Ogawa, S.; Iwasa, Y. *Nat. Mater.* **2004**, *3*, 317. (b) Pernstich, K. P.; Goldmann, C.; Krellner, C.; Oberhoff, D.; Gundlach, J.; Batlogg, B. *Synth. Met.* **2004**, *146*, 325.
- (26) (a) Pernstich, K. P.; Rashid, A. N.; Haas, S.; Schitter, G.; Oberhoff, D.; Goldmann, C.; Gundlach, D. J.; Batlogg, B. *Los Alamos Natl. Lab., Prepr. Arch., Condens. Matter* **2004**, 1–9, arXiv:cond-mat/0410014. (b) Kelley, T. W.; Boardman, L. D.; Dunbar, T. D.; Muiyres, D. V.; Pellerite, M. J.; Smith, T. P. *J. Phys. Chem. B* **2003**, *107*, 5877.
- (27) Kelley, T. W.; Muiyres, D. V.; Baude, P. F.; Smith, T. P.; Jones, T. D. *MRS Symp. Proc.* **2003**, *771*, 169.
- (28) (a) Deman, A. L.; Tardy, J. *Mater. Sci. Eng. C* **2006**, *26*, 421. (b) Itoh, E.; Miyairi, K. *Thin Solid Films* **2006**, *499*, 95. (c) Yoshida, M.; Uemura, S.; Kodzasa, T.; Kamata, T.; Matsuzawa, M.; Kawai, T. *Synth. Met.* **2003**, *137*, 967. (d) Liu, P.; Wu, Y.; Li, Y.; Ong, B. S.; Zhu, S. *J. Am. Chem. Soc.* **2006**, *128*, 4554.

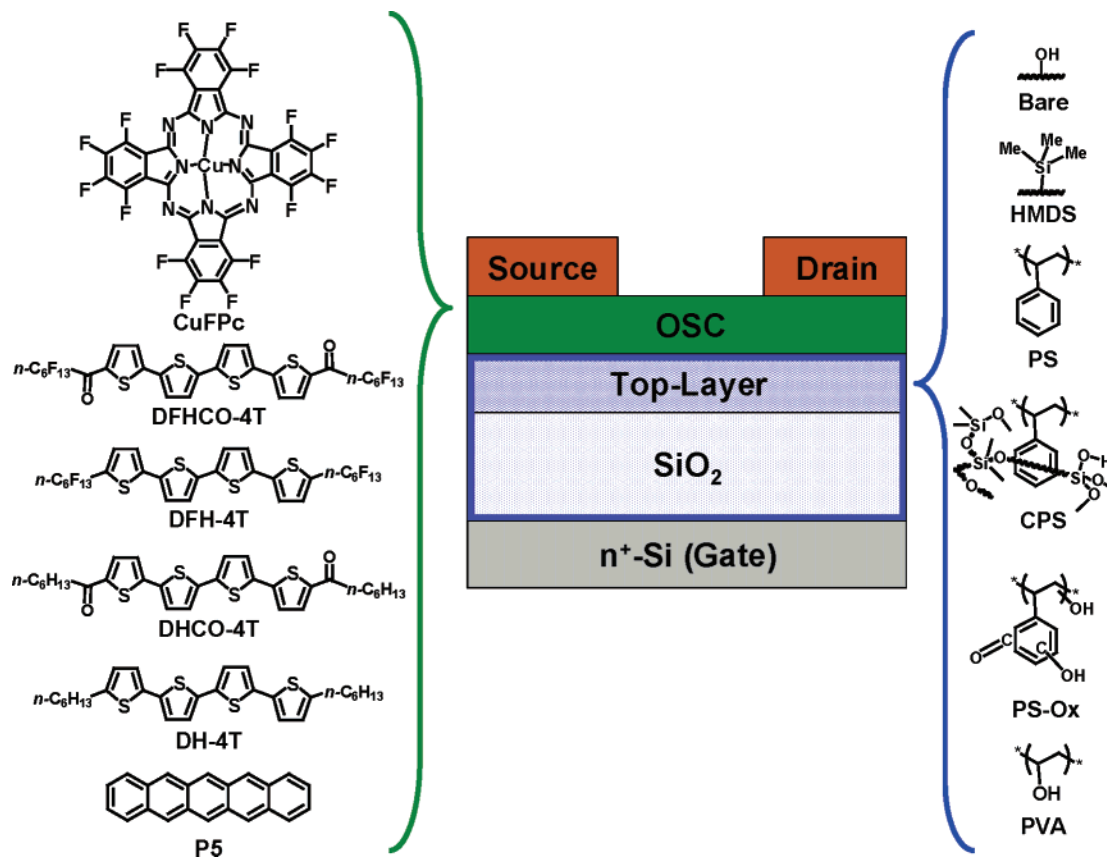


Figure 1. Chemical structure of the organic semiconductors (OSCs; left) and monolayers/polymers (right) employed in this study. The bottom gate OTFT device configuration is also shown (center).

interplay among SiO₂ dielectric root-mean-square (rms) roughness, pentacene morphology, and charge transport was further studied using variable-temperature charge transport measurements.²² In addition to dielectric morphology, the effects of dielectric surface chemical modification on device performance have been explored using self-assembled monolayers (SAMs).²³ In general, simple SiO₂ hydrocarbon functionalization using octadecyltrichlorosilane (OTS) or hexamethyldisilazane (HMDS) enhances the mobilities and lowers the off-currents of most OSCs.²⁴ More recently, it was shown that threshold voltage as well as mobility can be modulated by SAM dipole induced built-in surface potentials.²⁵ Although these results demonstrate the importance of controlling fundamental dielectric and/or interfacial properties for optimizing/controlling OSC charge transport efficiency, connections among dielectric bulk/surface chemistry, surface energy, dielectric-semiconductor interfacial morphology, and electrical properties are not completely clear,²⁶ particularly for far less developed electron-transporting organic semiconductors.

An attractive alternative approach for modifying dielectric surfaces is to deposit a second layer, such as a spin-on polymer, on top.²⁷ The major advantages of this approach are that, in contrast to SAM functionalization of oxide insulators, film deposition is not limited by the chemistry required for silane/phosphate coupling to the dielectric surface and different film thicknesses are readily accessed. Consequently, this methodology allows effective modification of bottom layer topography, hence planarization of rough dielectric surfaces, while simultaneously fine-tuning the chemical properties of the dielectric surface via choice of the deposited polymer. Despite this great

potential, to our knowledge investigations using this approach are sparse, limited to pentacene OTFTs. However, these suggest that pentacene carrier mobility can be tuned substantially, typically by a factor of 2–5×.²⁸ In this contribution, we report a systematic evaluation of the OTFT response characteristics of six organic semiconductors grown on four different SiO₂–polymer bilayer dielectric structures and compare them to HMDS-functionalized and pristine SiO₂ dielectrics. The questions we address are whether this bilayer dielectric strategy is a general approach to enhancing charge carrier mobility, whether there is a correlation between semiconductor–polymer combination and OTFT performance, and the nature of the dielectric surface chemical factors underlying such interfacial effects on charge transport. We demonstrate that OTFT mobility for certain semiconductors can be modulated to a very large extent (several orders of magnitude).

To address the above questions we selected for TFT fabrication OSCs with very different core structures, chemical functionalities, and frontier molecular orbital (FMO) energies, including those exhibiting hole, electron, and ambipolar transport on pristine/HMDS-treated SiO₂ (Figure 1).²⁹ The bottom layer of the bilayer dielectrics investigated in this study is 300-nm-thick SiO₂ thermally grown on p⁺-Si because of the ready accessibility of these substrates and the excellent insulating properties with extremely low leakage currents. Indeed, it has

(29) (a) Yoon, M.-H.; DiBenedetto, S. A.; Facchetti, A.; Marks, T. J. *J. Am. Chem. Soc.* **2005**, *127*, 1348. (b) Facchetti, A.; Mushrush, M.; Yoon, M.-H.; Hutchison, G. R.; Ratner, M. A.; Marks, T. J. *J. Am. Chem. Soc.* **2004**, *126*, 13859. (c) Facchetti, A.; Yoon, M.-H.; Stern, C. L.; Hutchison, G. R.; Ratner, M. A.; Marks, T. J. *J. Am. Chem. Soc.* **2004**, *126*, 13480. (d) Bao, Z.; Lovinger, A. J.; Brown, J. *J. Am. Chem. Soc.* **1998**, *120*, 207.

been reported that, in addition to the dielectric surface roughness¹⁸ and dielectric constant,³⁰ the leakage current through the gate insulator has major consequences for OTFT function.³¹ In this regard, a bilayer approach with SiO₂ should enable comparisons of the effects of various polymer surface functionalizations without complications of differing gate leakage currents through the different polymer dielectric layers.³² Furthermore, with this approach, transistor performance parameters on different dielectric modifications can be realistically compared over the same range of gate voltage/electric fields and charge densities accumulated at the semiconductor–dielectric interface, since the relatively thin polymer film can be employed on top of the SiO₂ and therefore maintain the total insulating layer thickness in a similar range. Note that polymeric dielectric materials typically require very different film thicknesses (sometimes very thick) to minimize gate leakage, and that it is not rare to observe mobility variations with the gate voltage/electric field associated with these thickness variations. In the present study, the top polymer layer was chosen/modified to provide a wide range of surface/film properties such as the chemical functionality, hydrophilicity, and polymer dielectric constant. Furthermore, the effect of using polymer layers of different thicknesses was investigated. All of the new bilayer dielectrics were characterized by impedance spectroscopy, quantitative leakage current density measurements, advancing aqueous contact angles, and atomic force microscopy (AFM). The effects of dielectric surface modifications on the OSC microstructure were investigated in detail using a combination of techniques including AFM, scanning electron microscopy (SEM), and wide-angle X-ray diffraction (WAXRD). It will be seen that polymer coating of inorganic insulators is a general strategy for strongly modulating electron transport in TFT devices, whereas hole transport is much less affected. This study thus demonstrates correlations between OTFT carrier mobility, relative semiconductor FMO energies, and the chemical nature of the dielectric surface. Insights into the chemical origin of major charge trapping sites are also provided, and it is shown that judicious choice of polymer coatings can “prime” any dielectric surface for organic semiconductor deposition to enhance OTFT performance.

Experimental Section

Materials. The semiconductors α,ω -diperfluorohexylcarbonylquaterthiophene (**DFHCO-4T**),^{29a} α,ω -dihexylcarbonylquaterthiophene (**DHCO-4T**),^{29a} α,ω -diperfluorohexylquaterthiophene (**DFH-4T**),^{29b} and α,ω -dihexylquaterthiophene (**DH-4T**)^{29c} were available in our laboratory from previous syntheses, while pentacene (**P5**) and hexadecafluorocopperphthalocyanine (**CuFPC**)^{29d} were purchased from Aldrich and purified by multiple gradient vacuum sublimation before use. Polystyrene (PS, $M_w = 280k$) and polyvinylalcohol (PVA, Mowiol 40-88, $M_w = 127k$) were purchased from Aldrich and used without further purification. Prime grade silicon wafers (p⁺-Si) with ~ 300 nm ($\pm 5\%$) thermally grown oxide (from Montco Semiconductors) were used as device substrates.

Film Deposition and Characterization. All p⁺-Si/SiO₂ substrates were cleaned by sonication in absolute ethanol for 3 min and then by

oxygen plasma treatment for 5 min (20 W). For the SiO₂ coating layer, –SiMe₃ groups were introduced using hexamethyldisilazane (HMDS), deposited by placing the SiO₂ substrates in an N₂-filled chamber saturated with HMDS vapor for 36–48 h. PS (5.0, 7.5, 15, or 30 mg/mL in anhydrous toluene), crosslinked-polystyrene³³ (CPS, 1:1 volume mixture of PS 7.0 mg/mL and 1,6-bis-trichlorosilylhexane in toluene), and PVA (30 mg/mL in Millipore water) were spin-coated onto substrates at 5000 rpm in the air (relative humidity $\sim 30\%$) and cured in a vacuum oven at 80 °C overnight. For **PS-Ox** coating, **PS1** films were exposed to an oxygen plasma for a minimal time (5 s, 20 W) before characterization and subsequent semiconductor deposition. Film thicknesses were measured by profilometry (Tencor, P10). Atomic force microscopic (AFM) images including rms roughness were obtained using a JEOL-5200 Scanning Probe Microscope with silicon cantilevers in the tapping mode, using WinSPM Software. For capacitance measurements, metal-insulator-semiconductor (MIS) structures were fabricated by depositing gold electrodes (200 $\mu\text{m} \times 200 \mu\text{m}$) on the polymer-coated p⁺-Si/SiO₂ substrates. All of the semiconducting materials were vacuum deposited at 2×10^{-6} Torr (~ 500 Å thickness, 0.2 Å/s growth rate) while maintaining the substrate temperature at ~ 50 °C. Thin films of organic semiconductors were analyzed by standard wide angle $\theta-2\theta$ X-ray film diffractometry (WAXRD) using monochromated Cu K α radiation. Semiconducting films were coated with 3 nm of sputtered Au before scanning electron microscopic (SEM) imaging using a Hitachi S4500 FE microscope. For FET device fabrication, top-contact electrodes (~ 50 nm) were deposited by evaporating gold (3×10^{-6} Torr) through a shadow mask with the channel length (L) and width (W) defined as 100 μm and 5000 μm , respectively.

Electrical Measurements. The capacitance of the bilayer dielectrics was measured on MIS structures using a Signaton probe station equipped with a digital capacitance meter (model 3000, GLK Instruments) and an HP4192A Impedance Analyzer. All OTFT measurements were carried out under a vacuum (1×10^{-5} Torr) using a Keithly 6430 subfemtoammeter and a Keithly 2400 source meter, operated by a local Labview program and GPIB communication. Triaxial and/or coaxial shielding was incorporated into the probe station to minimize the noise level. Mobilities (μ) were calculated in the saturation regime using the relationship³⁴: $\mu_{\text{sat}} = (2I_{\text{DS}}L)/[WC_i(V_G - V_T)^2]$, where I_{DS} is the source–drain saturation current; C_i is the gate dielectric capacitance (per area), V_G is the gate voltage, and V_T is the threshold voltage. The latter can be estimated as the x intercept of the linear section of the plot of V_G vs (I_{DS})^{1/2}.

Results

This account begins with a discussion of bilayer dielectric fabrication and surface characterization based on advancing aqueous angle measurements and AFM. For electrical characterization, leakage current and capacitance in MIS structures are then measured and discussed. Next, we describe the fabrication and characterization of OTFTs using various semiconductors and bilayer dielectrics. Finally, semiconductor growth mechanism and film morphologies/microstructures as probed by AFM, SEM, and WAXRD are discussed in the context of correlations between dielectric modifications and semiconductor film properties.

Bilayer Dielectric Fabrication and Characterization. All of the bilayer dielectric samples were fabricated on p⁺-Si/SiO₂ (300 nm) substrates. The top polymer layer was deposited by spin-coating according to the procedure described in the Experimental Section. The polymers employed in this study are

(30) (a) Veres, J.; Ogier, S. D.; Leeming, S. W.; Cupertino, D. C.; Khaffaf, S. M. *Adv. Func. Mater.* **2003**, *13*, 199. (b) Jang, Y.; Kim, D. H.; Park, Y. D.; Cho, J. H.; Hwang, M.; Cho, K. *Appl. Phys. Lett.* **2005**, *87*, 152105. (31) de Boer, R. W. I.; Iosad, N. N.; Stassen, A. F.; Klapwijk, T. M.; Morpurgo, A. F. *Appl. Phys. Lett.* **2005**, *86*, 032103/1. (32) Jang, Y.; Kim, D. H.; Park, Y. D.; Cho, J. H.; Hwang, M.; Cho, K. *Appl. Phys. Lett.* **2006**, *88*, 072101/1.

(33) Yoon, M.-H.; Yan, H.; Facchetti, A.; Marks, T. J. *J. Am. Chem. Soc.* **2005**, *127*, 10388.

(34) Sze, S. M. *Physics of Semiconductor Devices*, 2nd ed.; John Wiley & Sons: USA, 1981.

Table 1. Properties of the Bilayer Dielectrics Investigated in this Study Including Top-Layer Film Thickness (D , nm), Dielectric Surface RMS Roughness (ρ , nm), Advancing Aqueous Contact Angle (θ , deg), Effective Areal Capacitance (C_i , nF/cm²), Dielectric Constant (k_{top}), and Effective Dielectric Constant (k_{eff})

dielectric	D	ρ	θ	C_i	k_{top}	k_{eff}
BARE	0	0.1	<5	11.4	3.9	3.9
HMDS	0.4	0.1	102	11.4	2.8 ^a	3.9
PS1	24	0.2	92	10.3	2.5	3.8
PS2	31	0.2	92	10.0	2.5	3.7
PS3	71	0.3	92	8.5	2.5	3.6
PS4	150	0.3	92	6.5	2.5	3.3
PVA	115	0.3	45	9.5	7.4	4.5
CPS	13	0.9	40	10.7	2.6	3.7
PS-Oxy	24	0.3	<10	10.3	2.5	3.8

^a Estimated for PDMS polymers (ref 37).

polystyrene (PS), a crosslinked polystyrene blend (CPS),³³ and polyvinyl alcohol (PVA). Therefore, the following dielectric structures were fabricated/investigated and are identified here as the following (Figure 1, right): **Bare**, p⁺-Si/SiO₂(300 nm) treated with O₂ plasma before use; **HMDS**, p⁺-Si/SiO₂(300 nm) treated with HMDS vapor before use; **PS1**, p⁺-Si/SiO₂(300 nm)/PS(24 nm); **PS2**, p⁺-Si/SiO₂(300 nm)/PS(31 nm); **PS3**, p⁺-Si/SiO₂(300 nm)/PS(71 nm); **PS4**, p⁺-Si/SiO₂(300 nm)/PS(150 nm); **PS-Ox**, p⁺-Si/SiO₂(300 nm)/PS(24 nm) treated with an O₂ plasma; **CPS**, p⁺-Si/SiO₂(300 nm)/CPS(13 nm); **PVA**, p⁺-Si/SiO₂(300 nm)/PVA(115 nm). These samples allow investigating the effects of a wide range of surface energies, as indexed by advancing aqueous contact angle measurement data on the bilayer dielectrics (reported in Table 1), ranging from very hydrophilic (**Bare**, **PS-Ox**; $\theta < 10^\circ$), to moderately hydrophilic (**PVA**, **CPS**; $\theta \approx 40^\circ$), to hydrophobic (**PSn** and **HMDS**; $\theta > 90^\circ$). The samples offer a variety of surface chemistries and polymer thicknesses (**PS1-4**). Note that **PS-Ox**, which is prepared by exposing **PS1** to an oxygen plasma, exhibits essentially the same morphology and dielectric properties as those of **PS1**, but with a far more hydrophilic surface.³⁵ It will be shown that these modifications strongly affect the OTFT response for most of the organic semiconductors examined.

Typical leakage current densities of the surface-modified substrates are identical to that of pristine p⁺-Si/SiO₂ (**Bare**), $< 10^{-9}$ A/cm² at $E \approx 4$ MV/cm, as measured in MIS structures ($M = \text{Au}$, $200 \times 200 \mu\text{m}^2$ contact area). The insets of the AFM images in Figure 2 show that the current density versus voltage plots for the thinnest (**Bare**) and the thickest (**PS4**) insulators are identical. This result demonstrates that the leakage current densities at the maximum OTFT gate fields employed here (~ 3.3 MV/cm) are dominated by the bottom SiO₂ layer. AFM micrographs of the bilayer films reveal that, with the exception of **CPS** (rms roughness $\rho \approx 0.9$ nm), all dielectric samples exhibit very similar topographies characterized by very smooth AFM morphologies with $\rho = 0.1\text{--}0.3$ nm, slightly larger for the thicker **PSn** films (Table 1). Representative AFM images are also shown in Figure 2. Consequently, the differences among OTFT performance parameters (vide infra) can be mainly attributed to the chemical nature of the dielectric–semiconductor

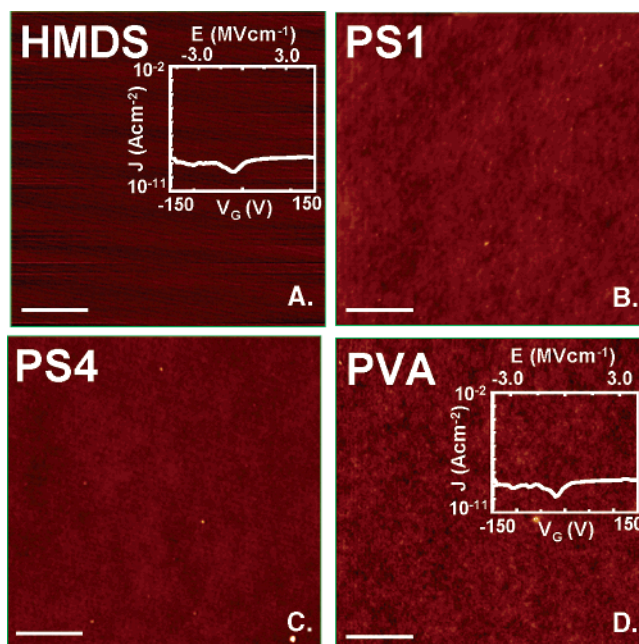


Figure 2. Tapping mode AFM images of: (A) **HMDS**, (B) spin-coated **PS1**, (C) spin-coated **PS4**, and (D) spin-coated **PVA** films on p⁺-Si/SiO₂ substrates. Insets show leakage current densities as a function of field for the indicated samples.

interface, since the dielectric surface roughness and gate leakage current are almost identical, regardless of the surface modification.

Table 1 also collects the areal capacitance ($\pm 5\%$) and the effective/top polymer layer dielectric constant data measured at 10 kHz for all dielectric samples. Capacitance–frequency plots (1–1000 kHz) shown in Figure 3 demonstrate that all dielectrics, with the exception of **PVA**, exhibit very little dispersion, typically $< 3\%$. The **Bare** and **HMDS** dielectrics exhibit the highest capacitance of 11.4 nF/cm², resulting in an effective dielectric constant of 3.9, identical to that reported in the literature for SiO₂.³⁶ A simple model of two parallel-plate capacitors in series is reasonably assumed to calculate the dielectric constant of the top polymer layer (k_{top}), and the relationship (the reciprocal additive rule) is depicted in the inset of Figure 3. The capacitance of **PSn** substrates gradually decreases from 10.3 (**PS1**) to 6.5 nF/cm² (**PS4**) when the top layer film thickness is increased from 24 to 150 nm. Note that the plot of PS layer thickness versus reciprocal bilayer capacitance is linear, with the y-intercept and slope providing the SiO₂ bottom layer capacitance ($1/0.087 = 11.5$ nF/cm²) and k_{top} of the top polystyrene layer ($k_{\text{top}} = 2.5$), respectively. From the effective capacitances of **PVA** (9.5 nF/cm²), **CPS** (10.7 nF/cm²), and **PS-Ox** (10.3 nF/cm²), the dielectric constants of the corresponding top PVA (7.4), crosslinked PS (2.6), and PS (2.5) polymer layers can be calculated. All values are very close to the bulk dielectric constants reported in the literature.³⁷

Thin-Film Transistor Fabrication and Characterization.

As discussed in the Introduction, studies on OTFTs fabricated with bilayer dielectrics (and most of those using a single polymer dielectric layer) have been limited to pentacene devices.^{21,22,28} With the goal of more fully understanding structure–property

(35) (a) Menard, M.; Paynter, R. W. *Surf. Interface Anal.* **2005**, *37*, 466. (b) Guruvanket, S.; Rao, G. M.; Komath, M.; Raichur, A. M. *Appl. Surf. Sci.* **2004**, *236*, 278. (c) Koprinarov, I.; Lippitz, A.; Friedrich, J. F.; Unger, W. E. S.; Woll, Ch. *Polymer* **1998**, *39*, 3001. (d) Murakami, T.; Kuroda, S.; Osawa, Z. *J. Colloid Interface Sci.* **1998**, *200*, 192.

(36) *Handbook of Chemistry and Physics*; CRC Press: R. C. West Ed.; Boca Raton, FL, 1988.

(37) Khastgir, D.; Adachi, K. *J. Poly. Sci. B* **1999**, *37*, 3065.

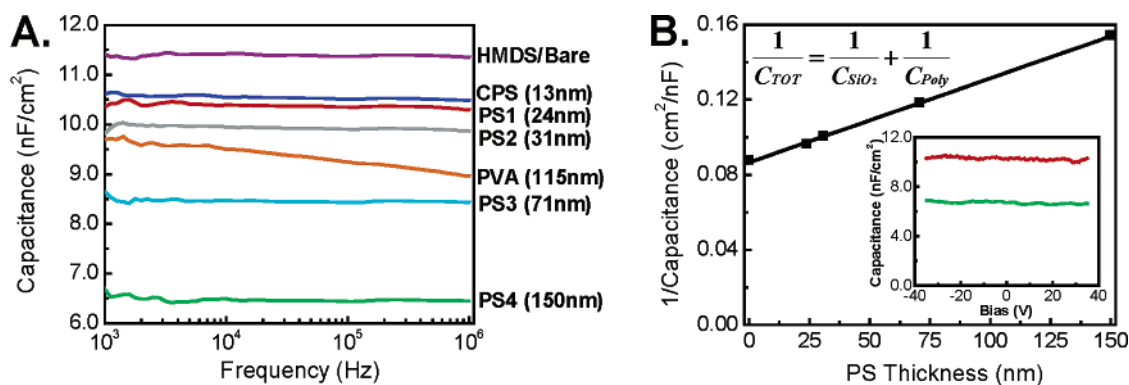


Figure 3. (A) Capacitance–frequency plots for bilayer p⁺-Si/SiO₂/polymer dielectrics measured on MIS structures (AC driving voltage = 0.1 V, DC bias offset = −10 V). (B) Inverse capacitance vs polystyrene top layer thickness plot. Inset: C_i vs bias plot for PS1 (red) and PS4 (green).

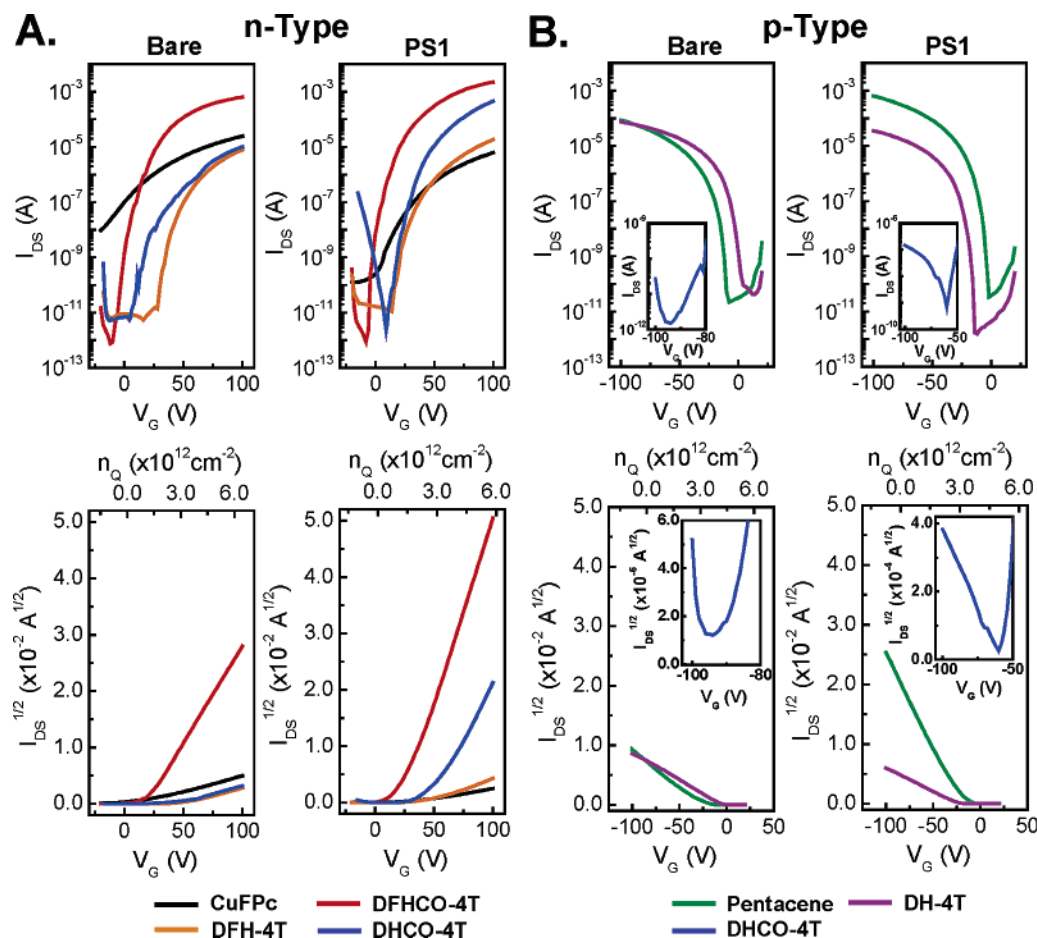


Figure 4. Comparison between the I_D versus V_G transfer plots (top, forward scan) and the corresponding square root of I_D versus V_G plots (bottom) for Bare- and PS1-based OTFTs for (A) n-channel ($V_G < 0$ V) and (B) p-channel ($V_G > 0$ V) OTFT operation.

relationships governing diverse organic semiconductor–dielectric interfaces, the OTFT performance characteristics of six semiconductors on nine bilayer dielectrics were analyzed. The semiconductors investigated here (Figure 1, left) were selected to span all possible combinations of majority carrier transport types observed on untreated/HMDS-functionalized SiO₂ dielectrics and are as follows: (i) N-type. Perfluoro-copperphthalocyanine (CuFpC), α,ω -diperfluorohexylcarbonyl-quaterthiophene (DFHCO-4T), and α,ω -diperfluorohexyl-quaterthiophene (DFH-4T); (ii) Ambipolar. α,ω -dihexylcarbonyl-quaterthiophene (DHCO-4T); (iii) P-type. α,ω -dihexylcarbonyl-quaterthiophene (DH-4T) and pentacene (P5) (Figure 1). Pentacene was included

since it has been widely investigated and can be used to compare our measurements to literature data on similar dielectric surfaces. These semiconductor molecular structures cover a broad selection in terms of majority carrier type, core architectural characteristics (oligothiophenes, phthalocyanine, and acene), and core substituent chemical functionalities (fluoroalkyl, alkyl, carbonyl, F, H). Furthermore, the intrinsic sensitivities of these semiconductors (especially n-type) to ambient conditions, primarily O₂ and H₂O vapor, are quite different suggesting different sensitivities to the dielectric surface chemistry and functionalities. Note that what is meant here by air sensitivity of an n-type semiconductor is not chemical reaction with air to

Table 2. OTFT Carrier Mobility (μ , $\text{cm}^2/(\text{V s})$), Current ON–OFF Ratio ($I_{\text{on}}/I_{\text{off}}$), Threshold Voltage (V_T , V) and Turn-ON Voltage (V_{on} , V) for Various Organic Semiconductors on Various Dielectrics

dielectrics	semiconductors ^a													
	n-type						ambipolar				p-type			
	CuFPC		DFHCO-4T ^b		DFH-4T		DHCO-4T (n)		DHCO-4T (p)		DH-4T		pentacene	
	μ ($I_{\text{on}}/I_{\text{off}}$)	V_T (V_{on})	μ ($I_{\text{on}}/I_{\text{off}}$)	V_T (V_{on})	μ ($I_{\text{on}}/I_{\text{off}}$)	V_T (V_{on})	μ ($I_{\text{on}}/I_{\text{off}}$)	V_T (V_{on})	μ ($I_{\text{on}}/I_{\text{off}}$)	V_T (V_{on})	μ ($I_{\text{on}}/I_{\text{off}}$)	V_T (V_{on})	μ ($I_{\text{on}}/I_{\text{off}}$)	V_T (V_{on})
BARE	0.010 (10 ⁴)	17 (−20)	0.44 (10 ⁸)	19 (−8)	0.001 (10 ⁶)	50 (28)	0.012 (10 ⁷)	50 (5)	7 × 10 ^{−8} (10)	−50 (−61)	0.022 (10 ⁶)	−9 (4)	0.12 (10 ⁷)	−30 (−8)
HMDS	0.011 (10 ⁶)	36 (4)	0.38 (10 ⁸)	21 (−6)	0.005 (10 ⁷)	60 (10)	0.22 (10 ⁷)	42 (14)	0.002 (10 ³)	−78 (−71)	0.015 (10 ⁶)	−1 (0)	0.14 (10 ⁶)	−23 (0)
PS1	0.007 (10 ⁴)	36 (6)	1.7 (10 ⁹)	24 (−6)	0.026 (10 ⁶)	48 (14)	0.70 (10 ⁸)	50 (12)	0.0003 (10 ²)	−60 (−58)	0.024 (10 ⁷)	−19 (−14)	0.43 (10 ⁷)	−24 (−2)
PS2	0.008 (10 ⁴)	30 (4)	1.5 (10 ⁹)	25 (−4)	0.024 (10 ⁶)	57 (20)	0.66 (10 ⁷)	51 (12)	0.0004 (10 ²)	−63 (−67)	0.025 (10 ⁷)	−20 (−14)	0.43 (10 ⁷)	−24 (−2)
PS3	0.006 (10 ⁴)	42 (8)	1.5 (10 ⁹)	35 (0)	0.025 (10 ⁶)	56 (18)	0.69 (10 ⁶)	53 (14)	0.0004 (10 ²)	−80 (−84)	0.025 (10 ⁷)	−18 (−6)	0.43 (10 ⁷)	−28 (−4)
PS4	0.005 (10 ⁴)	48 (10)	1.4 (10 ⁸)	47 (0)	0.009 (10 ⁶)	60 (24)	0.44 (10 ⁶)	61 (16)	0.0006 (10 ²)	−76 (−78)	0.025 (10 ⁷)	−18 (−4)	0.40 (10 ⁷)	−30 (−4)
PVA	0.0006 (10 ³)	21 (4)	1.1 (10 ⁸)	11 (−18)	0.002 (10 ⁶)	37 (4)	0.11 (10 ⁸)	40 (9)	0.0001 (10 ²)	−79 (−83)	0.017 (10 ⁷)	−31 (−22)	0.027 (10 ⁶)	−26 (−10)
CPS	0.007 (10 ⁴)	28 (4)	1.6 (10 ⁹)	22 (−10)	0.004 (10 ⁶)	60 (20)	0.30 (10 ⁷)	42 (10)	0.0002 (10 ²)	−48 (−28)	0.018 (10 ⁷)	−10 (−2)	0.22 (10 ⁷)	−17 (−4)
PS-Ox	0.006 (10 ³)	−31 (−36)	0.067 (10 ⁸)	23 (−8)	0.0002 (10 ⁶)	39 (32)	NA ^c NA	NA	0.0002 (10 ²)	−68 (−40)	0.016 (10 ⁵)	−14 (6)	0.24 (10 ⁵)	−24 (14)

^a The dielectric (gate) substrates were maintained at 50 °C during semiconductor film growth. ^b Deposited at room temperature. ^c NA means not active.

afford a new chemical species. Rather, as we and other groups have previously discussed,³⁸ physisorbed O₂/H₂O at grain boundaries causes electron trapping and suppression of the TFT activity. This is supported by the observation that such OTFT devices fully recover their activity if, after exposure to air, they are remeasured in a vacuum.

All of the present semiconductor films were grown by vapor deposition under a high vacuum ($\sim 10^{-6}$ Torr) while maintaining the substrate(gate)-insulator temperature at 50 °C. **DFHCO-4T** films were also deposited on bilayer dielectric substrates maintained at room temperature since we demonstrated previously that this semiconductor exhibits the greatest carrier mobility (on HMDS-treated SiO₂ dielectric) for this film deposition procedure.^{29a} Note that each semiconductor film deposition on the complete range of dielectric samples was performed in a *single batch* to avoid variations in film growth conditions. To assess reproducibility, two different bilayer batches corresponding to two separate monolayer/polymer preparations were used for semiconductor deposition. The final “top-contact” OTFT structures (Figure 1, center) were completed by thermal deposition of Au source/drain electrodes (50 nm thick, 200 × 5000 μm^2 wide), resulting in OTFT devices with a channel length (L) of 100 μm and a width (W) of 5000 μm . For each semiconductor, two OTFT device arrays, each containing 50 devices, were fabricated from each of the two dielectric batches. The devices were immediately transferred to a locally built vacuum probe station and maintained under a dynamic vacuum overnight before electrical characterization. The device exposure time to air (<5 min) was minimized to avoid environmental film doping/deep gas adsorption. All OTFT

measurements were performed under a vacuum ($< 10^{-5}$ Torr), and the $I_{\text{SD}}-V_{\text{G}}$ curves were analyzed using the standard metal-oxide-semiconductor field-effect transistor (MOSFET) model.³⁴ For each semiconductor and bilayer batch, at least 10 devices were measured, and no significant (<5%) variations were observed from device to device. Figure 4 shows typical I_{DS} vs V_{G} plots for all of the investigated semiconductors on **Bare** (untreated SiO₂) and **PS1** dielectrics. When a number of conditions are satisfied (e.g., $V_{\text{DS}} \geq V_{\text{G}}$), the channel becomes pinched and the source–drain current enters the saturation regime. The carrier mobility (μ_{sat}) and threshold voltage (V_T) can be calculated from the slope and the horizontal intercept of a linear part in $I_{\text{DS,sat}}^{1/2}$ vs V_{G} plot, respectively, according to eqs 1 and 2:

$$\mu_{\text{sat}} = \left(\frac{\partial \sqrt{I_{\text{DS}}}}{\partial V_{\text{G}}} \right)^2 \frac{2L}{WC_i} \quad (1)$$

$$V_{\text{T,sat}} = V_{\text{G}} - \sqrt{\frac{2I_{\text{DS}}L}{WC_i\mu}} \quad (2)$$

where C_i is the capacitance per unit area of the dielectric layer. Turn-on voltage (V_{ON}) is defined as the onset voltage at which I_{DS} begins to increase positively (n-type) or negatively (p-type). To correct for the effects of different dielectric thicknesses and dielectric constants, device carrier mobilities were calculated for the same range of accumulated charge carriers ($n_{\text{Q}} = C_i \cdot V_{\text{G}}/e = (4-5) \times 10^{12} \text{ cm}^{-2}$). In the following section, the effect of the bilayer dielectric structure on OSC TFT performance is described starting from the n-type semiconductors, followed by the ambipolar and p-type systems. Table 2 collects the OTFT performance parameters such as major carrier type, carrier mobility, threshold voltage, turn-on voltage, and current on–off ratio of all semiconductors studied for different bilayer dielectrics.

(38) (a) Chesterfield, R. J.; McKeen, J. C.; Newman, C. R.; Ewbank, P. C.; da Silva Filho, D. A.; Bredas, J.-L.; Miller, L. L.; Mann, K. R.; Frisbie, C. D. *J. Phys. Chem. B* **2004**, *108*, 19281. (b) Facchetti, A.; Yoon, M.-H.; Stern, C. L.; Katz, H. E.; Marks, T. J. *Angew. Chem., Int. Ed.* **2003**, *42*, 3900. (c) Facchetti, A.; Mushrush, M.; Katz, H. E.; Marks, T. J. *Adv. Mater.* **2003**, *15*, 33. (d) Katz, H. E.; Johnson, J.; Lovinger, A. J.; Li, W. *J. Am. Chem. Soc.* **2000**, *122*, 7787.

The first observation from the data in Table 2 is that all of the semiconductors exhibit the same operational polarity on all dielectrics, as was previously observed for HMDS-treated SiO₂ substrates.²⁹ Furthermore, we do not observe any of the n/p-type materials to exhibit ambipolar behavior induced by the dielectric, as reported for **P5** on PVA.^{17a} From these data it can be seen that the electron mobility of **CuFPc**, a well-known air-active n-type semiconductor, is ~ 0.01 cm²/(V s), almost insensitive to SiO₂ surface modification. The only exception is for the PVA-based **CuFPc** devices for which the mobility is about 1 order of magnitude lower than those with the other dielectrics. However, $I_{\text{on}}:I_{\text{off}}$ values change substantially, from $\sim 10^3$ on **PS-Ox** to as high as 10^6 on **HMDS**, indicating a variable degree of electron doping (vide infra) typical of this particular material.^{29d} In contrast, the carrier mobilities of the air-sensitive n-type materials are far more affected by the chemical nature of the dielectric surface. Hence, the OTFT performance parameters of partially air-stable^{29a} **DFHCO-4T** on **Bare** are similar to those on **HMDS** substrates with carrier mobilities of ~ 0.4 cm²/(V s) and $I_{\text{on}}:I_{\text{off}}$ of $\sim 10^8$. However, **DFHCO-4T** device parameters on the PS-modified SiO₂ and **CPS** dielectrics are significantly greater with the OTFT carrier mobility approaching 2 cm²/(V s). Interestingly, the off-currents are barely affected by the PS modification, resulting in a slightly increased current on–off ratio from 10^8 (**Bare**) to $> 10^9$ (**PSn**). To our knowledge, these n-type mobilities and current on–off ratios are comparable to the largest values reported to date for room-temperature OSC film growth and without mobility corrections for electrode contact resistance.^{2d} Compared to the **PS1**-based devices, significant diminution of transport characteristics is observed for PVA-based **DFHCO-4T** devices ($\mu \approx 1$ cm²/(V s); $I_{\text{on}}:I_{\text{off}}$ of $\sim 10^8$) and an even greater reduction is measured for **PS-Ox** devices ($\mu \approx 0.07$ cm²/(V s); $I_{\text{on}}:I_{\text{off}}$ of $\sim 10^8$). Similar μ and $I_{\text{on}}:I_{\text{off}}$ trends are observed when **DFHCO-4T** films are deposited at 50 °C (not shown in Table 2) although overall mobility values are lower by a factor of 2–3. Furthermore, similarly enhanced TFT performance is observed for devices fabricated with thicker PS coatings (**PS2–4**), demonstrating that the semiconductor–dielectric interface nature governs the carrier transport characteristics which are negligibly affected by the polymer coating layer thickness. This observation has significant implications for practical applications where it may be necessary to planarize relatively rough dielectric surfaces with polymer coatings. In contrast, more air-sensitive **DFH-4T**, an n-type semiconductor active only in a vacuum or an inert atmosphere,^{29b} exhibits the greatest sensitivity to the dielectric surface chemistry of the n-type semiconductors examined. Thus, the electron mobility of **DFH-4T** devices markedly increases from 0.001–0.002 cm²/(V s) on **Bare** and **PVA** to 0.004–0.005 cm²/(V s) on **HMDS** and **CPS** to 0.02–0.03 cm²/(V s) on **PSn**. As observed for the other semiconductors, **DFH-4T** devices fabricated with **PS-Ox** exhibit far lower mobilities ($\sim 10^{-4}$ cm²/(V s)). Note that all of the devices exhibit comparable off-currents ($\sim 10^{-11}$ A) and relatively high $I_{\text{on}}:I_{\text{off}}$ ratios of $\sim 10^6$ – 10^7 .

Bilayer dielectric-mediated charge transport variations are even more pronounced for organic transistors fabricated with **DHCO-4T**, which is one of the highest mobility ambipolar semiconductors discovered to date. We reported previously that optimized **DHCO-4T**-based TFTs on **HMDS** exhibit both p-

and n-type transport with carrier mobilities of 0.22 and 0.002 cm²/(V s), respectively.^{29a} Device n-channel operation performance varies dramatically from as low as being negligible on **PS-Ox** and poor on **Bare** ($\mu \approx 0.01$ cm²/(V s); $I_{\text{on}}:I_{\text{off}}$ of $\sim 10^7$) to excellent on **PSn** substrates ($\mu \approx 0.4$ – 0.7 cm²/(V s); $I_{\text{on}}:I_{\text{off}}$ of $\sim 10^6$ – 10^8), the latter values approaching those of the best **DFHCO-4T**-based OTFTs. Similarly to **DFHCO-4T** and **DFH-4T**-based TFTs, the electron mobility of **DHCO-4T** increases on proceeding from **PVA** (0.11 cm²/(V s)) to **CPS** (0.30 cm²/(V s)) substrates, while $I_{\text{on}}:I_{\text{off}}$ remains in the same range ($\sim 10^7$ – 10^8). Particularly interesting is the effect of the bilayer dielectric structure on the p-channel operation of **DHCO-4T** devices. For this borderline p-type material, TFT hole mobilities vary from as low as $\sim 10^{-7}$ cm²/(V s) on **Bare** to ~ 0.002 cm²/(V s) on **HMDS**, whereas, for the remaining bilayer dielectric structures, the values fall in the relatively narrow range (0.001–0.006 cm²/(V s)), regardless of the nature of the polymer modification. Note that, in contrast to n-channel operation, **PS-Ox** has no detrimental effect on **DHCO-4T** p-channel transport.

As far as p-type semiconductors are concerned, it is observed in the present work that the performance of **DH-4T**-based TFT devices is essentially invariant to the nature of the dielectric, whereas **P5** devices exhibit moderate variations of a magnitude in agreement with literature observations (vide infra).²² Thus, for all dielectric-**DH-4T** TFT combinations, the hole mobility is ~ 0.02 cm²/(V s) with $I_{\text{on}}:I_{\text{off}} \sim 10^7$. Pentacene devices on the same range of dielectrics respond in a slightly different manner than the **DH-4T** devices, with the carrier mobilities increasing from ~ 0.1 cm²/(V s) (**Bare** and **HMDS**) to ~ 0.2 cm²/(V s) (**CPS** and **PS-Oxy**) to ~ 0.4 cm²/(V s) on **PSn**. Remarkably, but fully understandably (vide infra), far lower mobilities are measured for PVA-based devices (~ 0.03 cm²/(V s)). The $I_{\text{on}}:I_{\text{off}}$ ratios of the **P5** devices vary from $\sim 10^5$ (**PS-Oxy**) to as high as 10^8 (**PS1**). Based on the TFT response characteristics of **DH-4T** and **P5** on the bilayer dielectrics, the effects of the dielectric surface modification on hole-transport properties using various polymeric films or **HMDS** are seen to be far less dramatic than the effects of the same range of modifications on “air-sensitive” electron-transporting TFT properties.

Another informative semiconductor–dielectric aspect illuminated by this study is the influence of the surface dielectric functionalization on the hysteresis of the $I_{\text{DS}}-V_{\text{G}}$ transfer characteristics, meaning the degree to which the I_{DS} current depends on the direction of the gate voltage sweep.³⁹ Although OTFT current–voltage hysteresis has potential applications in nonvolatile memory elements,⁴⁰ this phenomenon is detrimental to typical OTFT functions.⁴¹ $I_{\text{DS}}-V_{\text{G}}$ hysteresis has been ascribed to charge trapping in deep states⁴² and/or to dipole physical rearrangement/mobile ion accumulation at the dielectric–semiconductor interface.⁴³ The exact nature and chemical origin of these charged states has not been identified, especially in the case of polymeric insulators. Figure 5 shows representative transfer plots for both forward and reverse gate bias scans which demonstrate how surface SiO₂ modification affects $I_{\text{DS}}-V_{\text{G}}$

(39) Gu, G.; Kane, M. G.; Doty, J. E.; Firester, A. H. *Appl. Phys. Lett.* **2005**, *87*, 243512/1.

(40) Singh, Th. B.; Marjanovic, N.; Matt, G. J.; Sariciftci, N. S.; Schwodiauer, R.; Bauer, S. *Appl. Phys. Lett.* **2004**, *85*, 5409.

(41) Uemura, S.; Yoshida, M.; Hoshino, S.; Kodzasa, T.; Kamata, T. *Thin-Solid Films* **2003**, *438*–*439*, 378.

(42) Knipp, D.; Street, R. A.; Volkel, A.; Ho, J. *J. Appl. Phys.* **2003**, *93*, 347.

(43) Rep, D. B. A.; Morpurgo, A. F.; Sloof, W. G.; Klapwijk, T. M. *J. Appl. Phys.* **2003**, *93*, 2082.

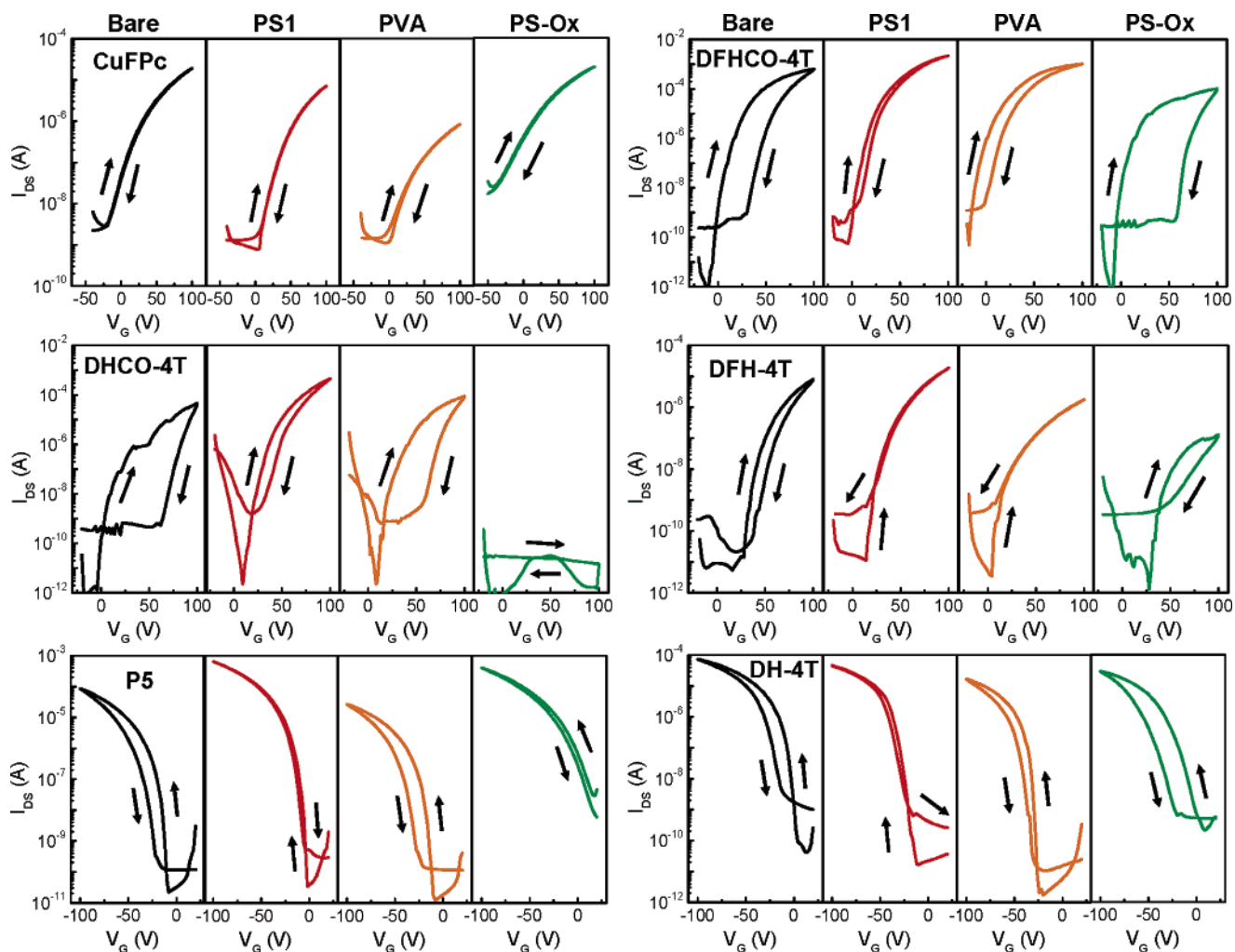


Figure 5. Comparison of forward and return I_{DS} versus V_G transfer plots for TFTs fabricated with the indicated OSC–dielectric combinations. Arrows denote gate bias sweep direction.

Table 3. Hysteresis (ΔV_G , V) and Subthreshold Voltage Swing (S , V/dec) Data for the Semiconductor–Dielectric Combinations Employed in this Study

dielectric	semiconductor											
	n-type						ambipolar		p-type			
	CuFpC		DFHCO-4T		DFH-4T		DHCO-4T ^a		pentacene		DH-4T	
	ΔV_G	S	ΔV_G	S	ΔV_G	S	ΔV_G	S	ΔV_G	S	ΔV_G	S
Bare	2	16.7	30	3.7	13	3.4	60	6.3	15	2.6	17	2.8
HMDS	2	9.1	28	3.2	30	3.2	25	1.8	3	2.0	6	2.4
PS1	1	5.6	7	1.5	2	1.7	12	1.6	3	1.6	5	2.0
PVA	5	10.9	15	2.9	2	4.3	38	2.8	18	2.3	12	2.7
CPS	2	5.6	14	1.3	12	3.8	30	2.9	5	1.2	10	2.7
PS-Ox	4	50.0	60	4.0	22	10.0	NA	NA	5	10.3	15	9.1

^a The hysteresis (ΔV_G) estimated by maximum gate voltage shift between the forward and backward sweep at a given I_{DS} ($V_{DS} = 100$ V) in I_{DS} vs V_G plot.

hysteresis. As a measure of hysteresis magnitude, we introduce the maximum gate voltage shift ($\Delta V_G = V_G^R - V_G^F$) at a given I_{DS} . The hysteresis data are collected in Table 3. From these data there are several informative trends to note. First, ΔV_G invariably exhibits the same sign, independent of the semiconductor–dielectric combination. Second, the extent of the ΔV_G change from that for the **Bare**-based TFTs is similar for all semiconductors, independent of the operation polarity, except for the **CuFpC** OTFTs. Third, the greatest ΔV_G variations are observed when comparing **PS1** versus **PS-Ox** devices, with the former exhibiting the smallest variations. Among the n-type

semiconductors, the **CuFpC**-based devices exhibit the lowest hysteresis with a maximum $\Delta V_G = 4$ –5 V on **PVA** and **PS-Ox**. Much larger I_{DS} – V_G hystereses are observed for n-type **DFHCO-4T** and **DFH-4T** and ambipolar **DHCO-4T** (n-channel operation) especially on **PS-Ox** ($\Delta V_G = 22$ –60 V), **Bare** ($\Delta V_G = 13$ –60 V), and **HMDS** ($\Delta V_G = 28$ –30 V). Typical hystereses for p-type semiconductors **P5** and **DH-4T** are generally lower than those observed for the air-sensitive n-type materials. The largest values for p-type semiconductors are measured for the **Bare** and **PVA** substrates ($\Delta V_G = 12$ –18 V), whereas hysteresis is strongly suppressed on **PS1** and **HMDS** ($\Delta V_G = 3$ –6 V).

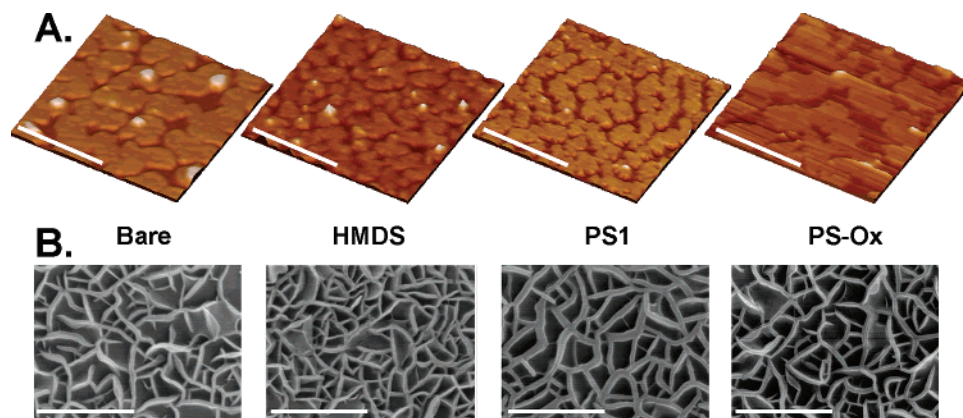


Figure 6. (A) Atomic force microscopic (~ 0.5 ML, ~ 2.0 nm thickness) and (B) scanning electron microscopic (~ 50 nm thickness) images of **DHCO-4T** films grown on the indicated gate dielectrics. Scale bars denote $1 \mu\text{m}$.

Substantial ΔV_G values are also observed for **DH-4T** on **CPS** and **PS-Ox** ($\Delta V_G = 10$ and 15 V, respectively).

Semiconductor Film Growth Mechanism and Film Microstructural Characteristics. Microstructural information is essential to understanding the origins of the observed dielectric-dependent TFT response variations. To address these issues, we must first understand if the largest mobility variations, primarily observed for the air-sensitive n-type materials, are due to differences in charge trapping within the semiconductor film or at the semiconductor–dielectric interfaces having different chemical properties or to a combination of both. The former should be largely governed by both: (i) intrinsic semiconductor molecular/film properties (FMO spatial and energetic characteristics, impurities, level of bulk molecular self-organization) which can reasonably be considered to be constant within the dielectric series, since the semiconductor films were grown simultaneously in the same batch, and (ii) the dielectric surface which should strongly influence the semiconductor film growth morphology (monolayer/bulk molecular ordering, crystallinity, density of nucleation sites and grain boundaries, molecular alignment). To a first approximation, the latter effects should be dominated by the dielectric characteristics and the processing history/conditions of the dielectric top surface, hence by bulk/interface chemical functionalities. Furthermore, in addition to TFT performance changes, for most of the oligothiophenes investigated here, the interesting question arises as to whether the film growth process is similar to that observed for widely studied pentacene, where different substrate-dependent nucleation kinetics induce dramatic variations in film morphology and microstructure.^{18–21}

The ambipolar semiconductor **DHCO-4T** exhibits the greatest substrate/dielectric-dependent electron mobility variations among the semiconductors investigated here, and the film growth properties will be discussed first. Figure 6 shows AFM and SEM images on four dielectrics for very thin (~ 2.0 nm nominal thickness, ~ 0.5 ML, ML = monolayer) and thick (~ 50 nm) vapor-deposited **DHCO-4T** films. The great similarity between the SEM pictures raises the possibility that thick film morphologies may not provide significant information on film growth mechanisms nor accurately represent **DHCO-4T** film morphologies in intimate contact with the dielectric surface. In contrast, the AFM images reveal that, for strongly hydrophilic, high surface energy dielectric surfaces such as **Bare** and **PS-Ox**, large **DHCO-4T** grains ($> 0.5 \mu\text{m}^2$) form on the dielectric surface,

whereas, for more hydrophobic substrates such as **PS1** and **HMDS**, crystallites with substantially reduced dimensions form ($0.1\text{--}0.2 \mu\text{m}^2$). It will be shown that, for some of the other semiconductors, even greater thin film morphological variations with dielectrics are observed (vide infra).

These results strongly suggest that different semiconductor film growth mechanisms are involved on different dielectric surfaces, which can be associated with formal Stranski–Krastanov and Volmen–Weber modes.⁴⁴ The former mechanism is invoked when molecule–molecule interactions are weaker than molecule–substrate interactions, whereas, in the latter, molecule–molecule interactions dominate. Note that a film growth characteristic of many oligothiophenes⁴⁵ is the tendency to form large (single) crystal plates extending along a crystallographic plane more or less orthogonal to the molecular long axes. This molecular arrangement maximizes core π – π stacking, which represents the dominant cohesive force in these molecular solids. Thus, the combination of these factors governs semiconductor film nucleation and evolution. In the initial film growth stage, **DHCO-4T** molecules impinging upon the **Bare** and **PS-Ox** substrates form nanoscopic nucleation sites that, due to the poor affinity of the hydrocarbon chains for the strongly hydrophilic surface, migrate and eventually coalesce laterally to form large grains. In contrast, in the case of hydrocarbon-functionalized **PS1** and **HMDS** dielectrics, every **DHCO-4T** nucleation site is sufficiently stabilized by the hydrophobic interaction with the dielectric surface to grow and form a large number of small crystallites. However, as argued by the SEM images, the **DHCO-4T** bulk film microstructures are practically identical for all substrates. To understand the origin of this result and the correlation between charge transport in OTFT semiconducting layers and morphological differences near the semiconductor–dielectric interface, **DHCO-4T** film growth on the various bilayer dielectric layers was sequentially monitored from ~ 0.5 ML to ~ 2 ML by AFM. That TFT charge transport is confined in close proximity to the dielectric interface has been suggested,⁴⁶ and Dinelli for α -sexithiophene^{47a} and Muck for

(44) Smith, D. L. *Thin-Film Deposition: Principles and Practice*; McGraw Hill: New York, 1995; Chapter 5.

(45) Fichou, D. *J. Mater. Chem.* **2003**, *10*, 571 and references therein.

(46) Dodabalapur, A.; Torsi, L.; Katz, H. E. *Science* **1995**, *268*, 270.

(47) (a) Dinelli, F.; Murgia, M.; Levy, P.; Cavallini, M.; Biscarini, F. *Phys. Rev. Lett.* **2004**, *92*, 116802. (b) Muck, T.; Wagner, V.; Bass, M.; Leufgen, J.; Geurts, J.; Molenkamp, L. W. *Synth. Met.* **2004**, *146*, 317. (c) Ruiz, R.; Papadimitratos, A.; Mayer, A. C.; Malliaras, G. G. *Adv. Mater.* **2005**, *17*, 1795.

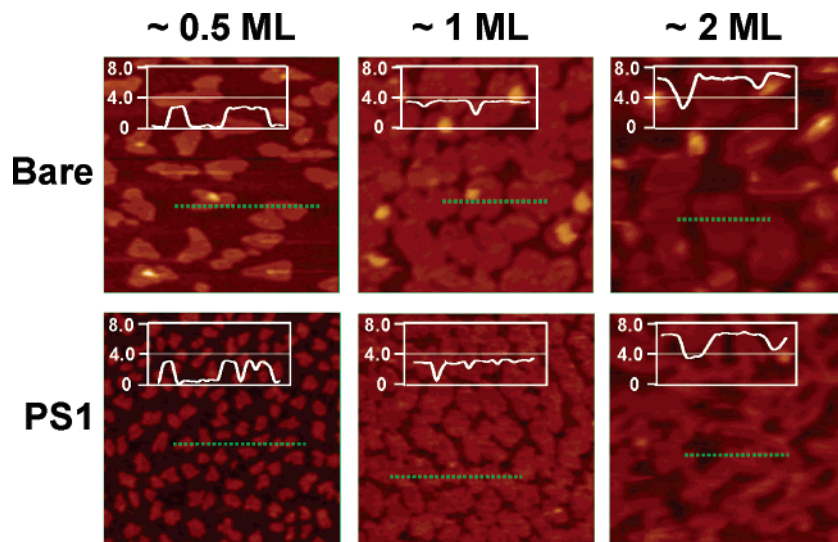


Figure 7. Atomic force microscopic images of **DHCO-4T** films of different thicknesses grown on **Bare** and **PS1** substrates. The insets show height profiles across indicated portions of the surface. All scan areas are $2 \mu\text{m} \times 2 \mu\text{m}$.

dihexylquaterthiophene^{47b} clearly demonstrated that the first two molecular layers on the dielectric dominate TFT carrier mobility. More recently Malliaras has shown that carrier mobility saturates at about six monolayers for pentacene TFTs.^{47c}

Figure 7 (left) shows AFM images of ~ 0.5 ML **DHCO-4T** films on **Bare** (large grains) and **PS1** (small grains). With increasing film thickness, both small and large grains on **Bare** and **PS1** samples, respectively, eventually coalesce to form generally uniform first and second layers before the onset of bulk film growth. Interestingly, despite the presence of the perfluoroalkyl chains, a very similar growth process is observed for **DFHCO-4T** as compared to **DHCO-4T** (See Figures S1 and S2). These growth mode data are further supported by WAXRD experiments (vide infra). Therefore, the principal differences in **DHCO-4T** and **DFHCO-4T** TFT performance on the different dielectrics is not dominated by different film growth modes or interfacial morphological variations on the dielectric surface since *all the dielectrics afford very similar semiconductor film microstructures* for thicknesses > 6 nm (~ 2 ML) of interest for TFT transport.

Particularly instructive is the comparison of the first few layers of growth/morphology for perfluorohexyl-substituted quaterthiophene (**DFH-4T**) versus the corresponding alkyl-substituted quaterthiophene (**DH-4T**), as shown in Figure 8. The former material is an n-type semiconductor, and its TFT performance is strongly affected by the dielectric surface modifications, whereas the latter is a p-type semiconductor and its TFT performance is essentially invariant to the dielectric surface modification. The bulk morphologies of **DFH-4T** and **DH-4T** are quite different (see SEM images in Figures 8, S3, and S4). The former is characterized by well-interconnected elongated grains, while the latter exhibits very large crystallites separated by deep channels. For each semiconductor, similar morphologies are observed for the thick films on the remaining dielectrics (not shown). It is anticipated that the fluorocarbon chains of **DFH-4T** will have poor affinity for both hydrophilic and hydrocarbon-functionalized surfaces. Therefore, submonolayer **DFH-4T** films on all dielectrics are characterized by very large two-dimensional plates spanning several microns. Interestingly, the largest and most continuous submonolayer grains (> 1

μm^2) are formed on very hydrophilic **Bare** and **PS-Ox** substrates, where the corresponding **DFH-4T** TFTs exhibit the lowest carrier mobilities. However, in contrast to the previous semiconductors and **DH-4T** (vide infra), the **DFH-4T** submonolayer to multilayer transition (shown in Figure 8A for **PS1**) occurs before the first layer is completely filled. Note that a very similar film growth pattern is observed on all the other dielectrics, demonstrating again that the large electrical parameter variations in **DFH-4T** OTFTs are *not* due simply to different semiconductor film morphologies. From a completely different perspective, the same conclusions can be drawn when analyzing **DH-4T** film growth. Note that it has been reported previously that ultrathin **DH-4T** films tend to grow two-dimensionally on SiO_2 with very uniform and large grains, and that the first two monolayers of this particular semiconducting material dominate field-effect mobility in the bottom-contact TFT configuration.⁴⁸ In contrast to **DFH-4T**, very large differences in the initial film growth pattern are observed for **DH-4T**, the details of which depend on the particular bilayer structure (see Figure 8B). However, all films on the different bilayer structures exhibit similar morphological transitions from submonolayer to bilayer to bulk film (shown in Figure 8-bottom for **PS1**). Note that the dielectric surface dependent differences between the initial film growth mechanisms (up to ~ 2 ML) are the greatest for **DH-4T** among the oligothiophenes investigated. However, note that the hole mobility for **DH-4T** is *completely unaffected*, again arguing for predominant molecular control of the OTFT parameters for this particular semiconductor (vide infra).

Some aspects of the influence of dielectric surface functionalization, as well as other film deposition parameters, on the morphology of pentacene and **CuPc** films were recently reported.¹⁷ For these systems, we find morphological variations in the present study that are similar to the previous reports with some informative exceptions. Typically, when submonolayer **P5** films exhibit a large (small) number of nucleation sites, this invariably gives rise to the formation of small (large) grains for the bulk films with more (less) grain boundaries.²⁰ Figure

(48) Muck, T.; Fritz, J.; Wagner, V. *Appl. Phys. Lett.* **2005**, *86*, 232101-1.

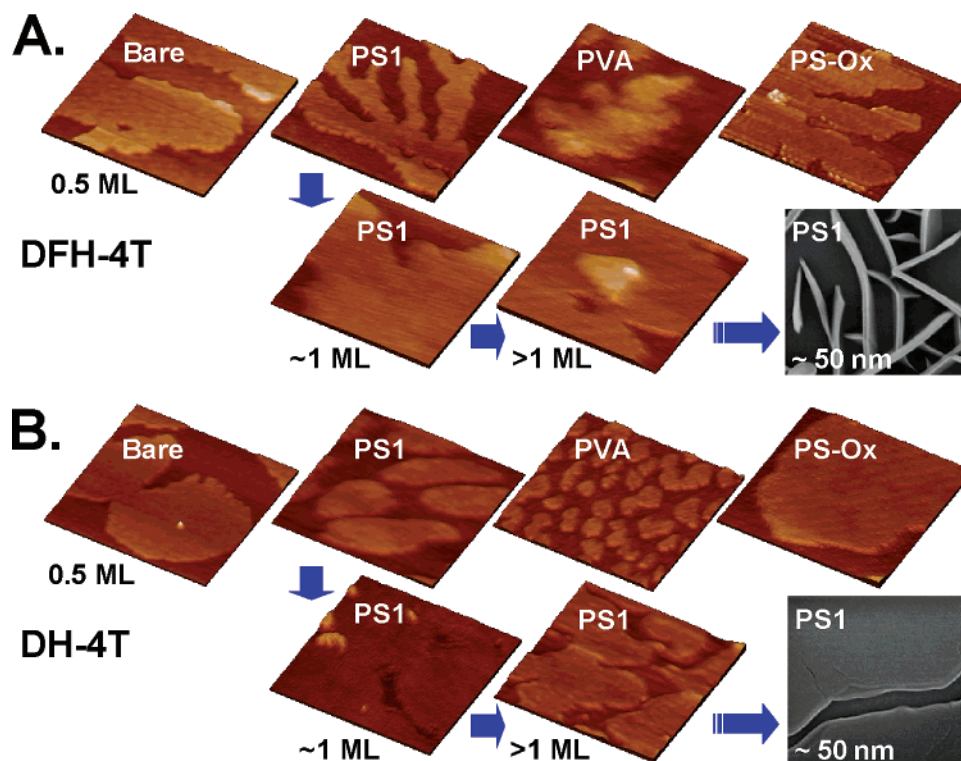


Figure 8. AFM (0.5–~1.5 monolayer) and SEM (50 nm thick) images of (A) **DFH-4T** and (B) **DH-4T** films on the different dielectric substrates. Images in the lower row represent semiconducting films thicker than 0.5 monolayer. All scan areas are $2\ \mu\text{m} \times 2\ \mu\text{m}$.

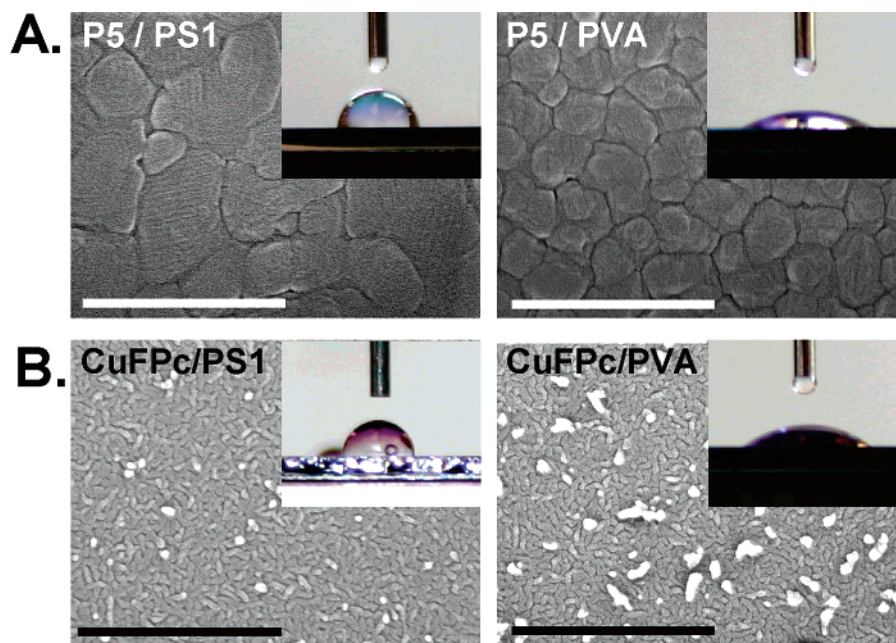


Figure 9. Scanning electron microscopic images of (A) **P5** (pentacene) films on **PS1** and **PVA** and (B) **CuFPc** films on **PS1** and **PVA**. Scale bar denotes $1\ \mu\text{m}$. Insets are images of water drops on semiconducting films for contact angle measurement.

9A shows SEM images for 50 nm thick **P5** films deposited on **PS1** and **PVA**, which indicate that this material exhibits the greatest dielectric surface-induced bulk morphological changes within the present semiconductor series (see also Figures S5 and S6). Although all thick **P5** films exhibit the formation of typical terraced features, the grain size on **HMDS** (not shown) and on **PS1** substrates is much larger ($\sim 7\times$ and $5\times$, respectively) than those observed on **PVA**. A plausible explanation for the large grains on PS is the affinity of pentacene for an

acene-functionalized surface such as polystyrene. A similar phenomenon has been reported for pentacene TFTs fabricated on a phenyl-containing monolayer gate dielectric for which good pentacene self-organization and mobility were measured.⁴⁹ N-channel **CuFPc** films grown at relatively low temperature ($<100\ ^\circ\text{C}$) are invariably characterized by very small crystal-

(49) Halik, M.; Klauk, H.; Zschieschang, U.; Schmid, G.; Demh, C.; Schutz, M.; Maisch, S.; Effenberger, F.; Brunnbauer, M.; Stellacci, F. *Nature* **2004**, *431*, 963.

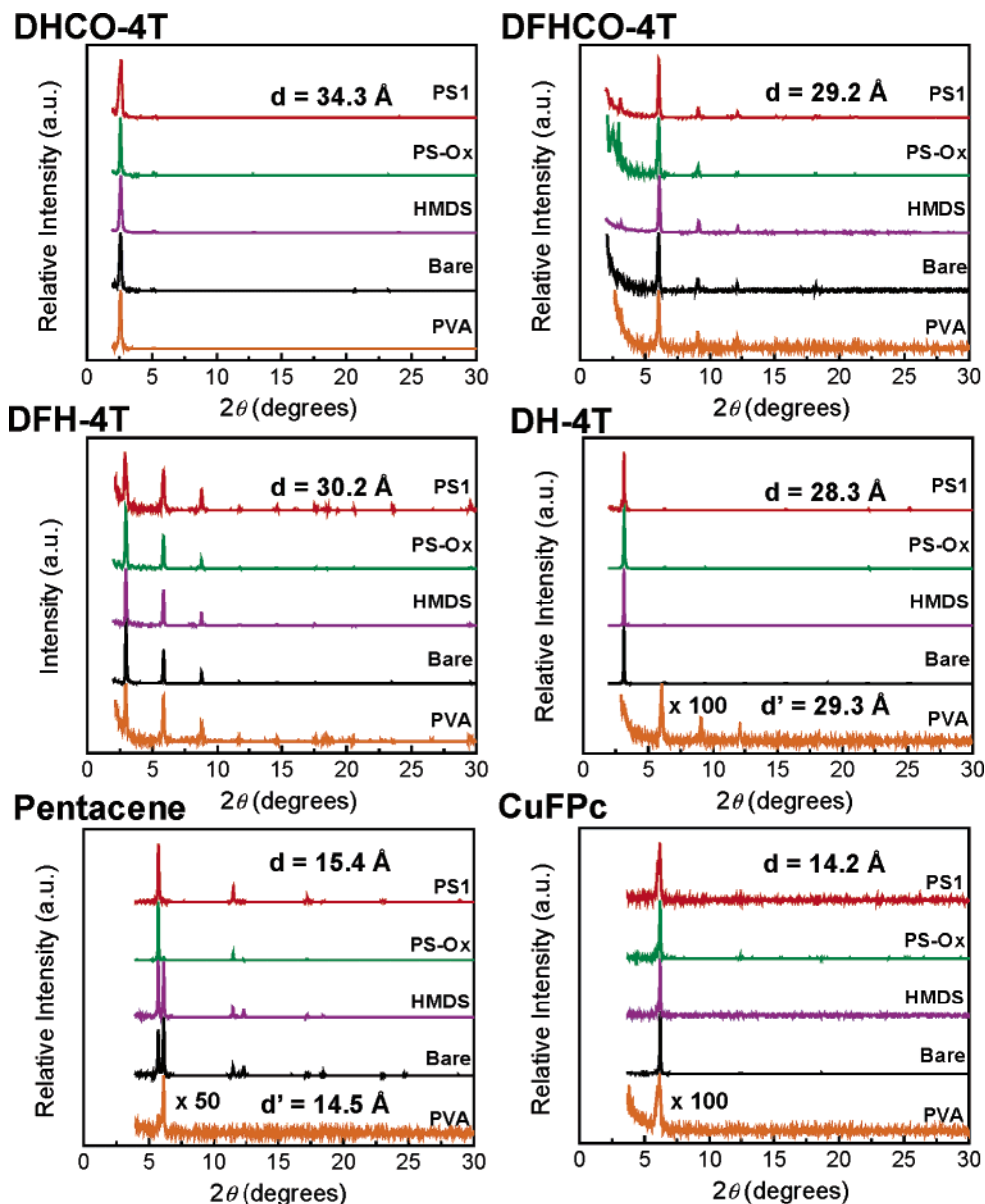


Figure 10. WAXRD θ – 2θ scans for the indicated organic semiconductor (50 nm thick)–bilayer dielectric combinations.

lites.⁵⁰ Similar results are observed here for **CuFPc** films deposited on the other bilayer dielectrics. The greatest crystallite size variations, still relatively small compared to **P5** films, are seen again when comparing **HMDS** (not shown) and **PS1** dielectrics/substrates to **PVA**. **CuFPc** films on the latter dielectric exhibit the formation of a large number of flakes with 2–3 \times smaller grain size (Figure 9B-right).

Equally substantial evidence for dramatically different **P5** and **CuFPc** film morphologies on **PVA** compared to those on **PS1** is revealed by advancing aqueous contact angle measurements on surfaces of the semiconductor films. As shown in Figure 9, the wettability of **P5** films is very different on going from **PS1** (as well as the other dielectrics) to **PVA** substrates, since θ changes from $\sim 85^\circ$ to $\sim 20^\circ$. **CuFPc** behaves similarly, with θ being $\sim 95^\circ$ for all dielectrics with the exception of **PVA** ($\theta \sim 25^\circ$). Furthermore, θ decreases even more with time, and

eventually the water drops spread completely and delaminate the semiconductor films. This is a clear demonstration that water can easily penetrate between the **P5** or **CuFPc** grains and dissolve the **PVA** coating beneath. Hence, the channels between grains are deep and reach the **PVA** surface. Note that this is not a peculiarity of all the **PVA**/semiconductor structures (**PVA** is water-soluble), since, in the case of the semiconductor films other than pentacene and **CuFPc**, the contact angle is essentially independent of the underlying dielectric layer and time and is found to be as follows: $\sim 90^\circ$ (**DH-4T**), $\sim 100^\circ$ (**DHCO-4T**), $\sim 110^\circ$ (**DFH-4T**), $\sim 130^\circ$ (**DFHCO-4T**). Since water cannot reach the dielectric surface, this is clear evidence that, for all of the present oligothiophene semiconductors, all bilayer dielectric surfaces are covered by (at least) a completely filled molecular layer.

To further investigate film semiconductor microstructure and degrees of texture, WAXRD measurements were performed for all semiconductor–dielectric combinations. Figure 10 shows θ – 2θ scans and the corresponding d -spacings. With the

(50) (a) Oh, Y.; Pyo, S.; Yi, M.-H.; Kwon, S.-K. *Org. Electron.* **2006**, *7*, 77. (b) Basova, T.; Kol'tsov, E.; Hassan, A.; Tsargorodskaya, A.; Ray, A.; Igumenov, I. *Phys. Stat. Sol.* **2005**, *242*, 822.

exception of pentacene, all of the organic semiconductor films on the different dielectrics exhibit a single progression of equally spaced Bragg reflections. The d -spacing patterns of the ambipolar/n-type materials [**DHCO-4T** (34.3 Å), **DFHCO-4T** (29.2 Å), **DFH-4T** (30.2 Å), and **CuFPC** (14.2 Å)] are completely insensitive to the dielectric functionalization and are identical to the values observed for HMDS-treated SiO₂ substrates.²⁹ For all semiconducting films in the present study, a general trend has been observed in the maximum intensities of each semiconductor film (of identical thickness) WAXRD scan on the various bilayer dielectrics falls in the order of **PS-Ox** > **Bare** \approx **HMDS** > **PS** \approx **PVA**. Importantly, the highest and lowest degrees of texture in air-sensitive n-type semiconductor films on **PS-Ox** and **PS/PVA**, respectively, are not directly correlated with the apparent device performance trends (Table 2, $\mu_{\text{PVA}} \gg \mu_{\text{PS-Ox}}$). Considering that for the air-sensitive n-type semiconductors, each type of semiconductor film on the various bilayer structures is prepared in a single batch leading to the same film thickness, and that the maximum θ - 2θ scan intensity variation on different bilayer structures is relatively small (within a factor of 2–5), this provides additional evidence that their transistor performance variation can be ascribed to differences in semiconductor–dielectric interface chemical properties rather than to the film morphology/microstructures (vide infra). In contrast, the intensity of the WAXRD scans of the air-stable n-type and p-type semiconductors on **PVA** is significantly lower by a factor of 50–100 than those on the other dielectrics as indicated in Figure 10, while the overall maximum intensity variation on various bilayer dielectrics follows the trend **PS-Ox** > **Bare** \geq **HMDS** > **PS** \gg **PVA**. Especially, in the case of **CuFPC** on **PVA**, the first-order diffraction peak exhibits not only substantially less intensity but also a twice larger full width at half-maximum (fwhm \approx 0.4° in 2θ) than the same semiconductor films on the other bilayer dielectrics, indicating a poorly ordered film microstructure on **PVA**. The poor crystallinity of **CuFPC**, **DH-4T**, and **P5** films on **PVA** as revealed by WAXRD can be correlated with the relatively poor performance of the corresponding TFT devices (Table 2).

In addition to the relatively low degrees of film texturing, **DH-4T** and **P5** films on **PVA** exhibit different molecular orientations from those observed on the other dielectric surfaces. **DH-4T** WAXRD scans exhibit a single set of reflections; however, the d -spacing calculated for films grown on **PVA** (29.3 Å) is significantly larger than that found on all of the other substrates (28.3 Å), demonstrating a different growth mode. WAXRD θ - 2θ scans of the **P5** films reveal an interesting dielectric-promoted microstructural transition, the details of which will be discussed elsewhere.⁵¹ Briefly, **P5** films on **PS1** and **PS-Ox** are characterized by an almost phase-pure film, with a d -spacing of 15.4 Å, whereas on **PVA** evidence of a different single phase is observed with $d = 14.5$ Å, and these two d -spacings correspond to the so-called, previously identified “thin film” and “bulk” pentacene phases, respectively.^{18a} Note that the WAXRD scans of **P5** films on **HMDS** and **Bare** exhibit the presence of both phases with comparable diffraction intensities. To a greater extent, the films of the p-type materials on **PVA** are characterized by Bragg progressions with far smaller intensities and broader widths than those of the same

thickness grown on the other substrates, indicating that both **P5** and **DH-4T** on **PVA** exhibit less ordered film microstructures.

To summarize these observations, the morphology and microstructure within the air-sensitive n-type/ambipolar semiconductors are rather insensitive to the dielectric layer surface, whereas the TFT charge transport is extremely sensitive. In contrast, the air stable n-type and p-type materials exhibit relatively modest dielectric-related TFT performance alterations despite the much greater variations in semiconductor film morphology, crystallinity, and molecular orientation.

Discussion

The present results for OTFTs fabricated on bilayer dielectrics having diverse surface chemical functionalities and using a broad range of organic semiconductor types provide new insights into critical relationships between OTFT performance parameters and semiconductor molecular/electronic structure, semiconductor film microstructure evolution, growth mechanism, and dielectric surface characteristics. Although it is not always possible to unambiguously differentiate among the relative contributions of all of these interconnected effects, this investigation provides clear evidence as to which of the aforementioned effects dominates for devices fabricated with most semiconductor–dielectric combinations. In the following discussion we focus on relationships between OTFT performance and dielectric surface chemistry.

Since unfunctionalized SiO₂ is the substrate from which all of the functionalized/bilayer dielectrics are fabricated, we use this insulator as a reference point to better understand the origins of performance variations. In the case of carrier mobility, to clearly visualize changes, we introduce the enhancement factor $\eta = \mu_{\text{X}}/\mu_{\text{Bare}}$ defined as the ratio of the field-effect carrier mobility observed for a certain bilayer structure **X** (μ_{X}) and that measured on the **Bare** (μ_{Bare}) substrate. These results are plotted in Figure 11A for all of the semiconductors investigated. It can be clearly seen that η increases on average when moving from the sides to the center of the plot, hence from the less air-insensitive n-type and p-type semiconductors to the more sensitive n-type and ambipolar materials. Therefore, there is a distinct correlation between the dielectric surface chemistry and the empirical sensitivity of the semiconductor majority carrier type to ambient. Figure 11B plots the electrochemically derived frontier molecular orbital energy levels (HOMO and LUMO) for the semiconductors employed in this study. Note that the low-lying LUMO and high-lying HOMO molecules are generally those exhibiting the least sensitivity to the nature of the dielectric surface. Consequently, the utility of employing OSCs having progressive variations in MO energies to probe semiconductor–dielectric interfacial properties finds experimental confirmation.

The general trend in OTFT mobilities as a function of the various bilayer dielectrics can be summarized by the following observations: (1) For air-stable n-type semiconductors such as **CuFPC** as well as cyanated perylene derivatives,⁵² OTFT performance parameters are relatively insensitive to the dielectric

(51) Kim, C.; Facchetti, A.; Marks, T. J. Manuscript in preparation.

(52) We have synthesized [Jones, B.; Ahrens, M. J.; Yoon, M.-H.; Facchetti, A.; Marks, T. J.; Wasielewsky, M. R. *Angew. Chem., Int. Ed.* **2004**, *43*, 6363] and investigated a number of alkyl-substituted cyanated perylenes and found very little change in carrier mobility on different dielectrics. Jones, B.; Wasielewsky, M.; Marks, T. J. Manuscript in preparation.

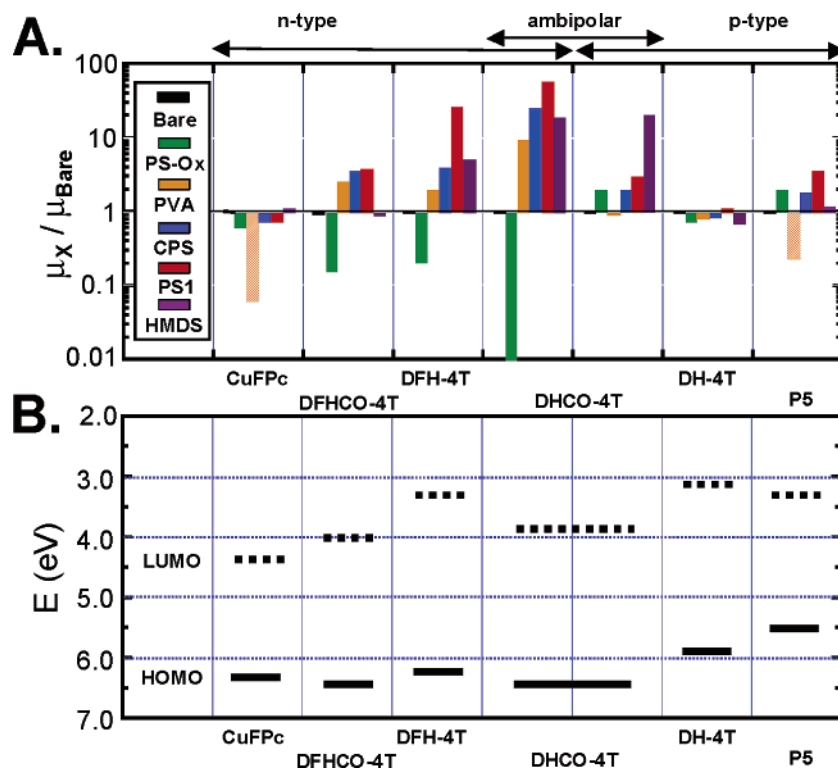


Figure 11. (A) Histogram showing the mobility enhancement ratio $\eta = \mu_x / \mu_{\text{Bare}}$ and (B) Electrochemically derived FMO energy levels for the organic semiconductors investigated in this study. The crosshatched bars denote semiconductor–dielectric combinations for which the largest variations in semiconductor film morphology are observed.

surface ($0.7 < \eta_{\text{CuFPc}} < 1.1$), with the reasonable exception of PVA (vide infra). (2) The n-type mobilities and current on–off ratios of air-sensitive n-type semiconductors such as **DFHCO-4T**, **DFH-4T**, and ambipolar **DHCO-4T** vary substantially ($0 < \eta_{\text{n-type}} < 100$) with the nature of dielectric surface, and the performance enhancement is most pronounced in the n-type mobility of **DHCO-4T** films on **PSn** dielectrics. (3) The effects of bilayer dielectrics on the p-type semiconducting properties of **DH-4T** and **P5** are minimal ($0.8 < \eta_{\text{p-type}} < 1.3$, with the exception of **PVA-P5** where $\eta_{\text{P5}} = 0.2$) even on **PSn** and **PS-Ox**, which induce dramatic mobility increases and decreases, respectively, in the air-sensitive n-type devices.

In the present study, considering that each semiconductor is simultaneously grown on the various bilayer dielectric layers in a single batch and that great similarities have been demonstrated in semiconductor film morphologies and microstructures, it is reasonable that the observed performance differences for each semiconductor set can be largely attributed to differences in the details of the dielectric–semiconductor interface chemistry. As shown from variable-temperature mobility studies on n-channel organic transistors, the charge transport in these materials is largely limited by poorly understood charge traps,³⁸ in contrast to the coherent bandlike transport operative in most inorganic semiconductors. These traps are thought to be localized at chemical/physical defects, at semiconductor grain boundaries, and/or at the semiconductor–dielectric interface.⁵³ Trap density changes can be estimated by relative V_T shifts with respect to that in reference samples ($\Delta n_{\text{trap}} = (C_i \Delta V_T) / q$),⁵⁴

where q is the charge on an electron, or by the difference between V_T and V_{on} in the same device ($n_{\text{trap}} = [C_i(V_T - V_{\text{on}})] / q$).⁵⁵ Note that V_T is a fitting parameter derived from $(I_{\text{DS}})^{1/2}$ vs V_G plots and can vary substantially, depending on the applied gate bias,⁵⁶ especially when OTFT devices exhibit gate-bias dependent mobility, gate stress effects, and/or hysteresis. In contrast, the subthreshold swing (S) should be less dependent on the aforementioned artifacts.⁵⁷ Regarding the electrical properties of dielectric–semiconductor interfaces in TFT devices having different dielectric structures, the maximum density of traps can be estimated from eq 3:⁵⁸

$$N_{\text{trap}}^{\text{max}} = \frac{C_i}{q} \left[qS \log e - 1 \right] \quad (3)$$

where k_B is Boltzmann's constant, T is temperature, e is the base of the natural logarithm, and C_i is the areal capacitance of the dielectric structure. The estimated trap densities for the various semiconductor dielectric pairs are depicted in Figure 12. When bulk trap densities in the semiconductor layer are similar among different dielectrics or are far less than interface trap densities, the trend in trap density estimated from eq 3 can be ascribed mainly to trapped charges at the semiconductor–dielectric interface and used as the maximum estimated interface trap density.^{57b} In the present study, this is a valid assumption

(53) Newman, C. R.; Frisbie, C. D.; da Silva Filho, D. A.; Bredas, J.-L.; Ewbank, P. C.; Mann, K. R. *Chem. Mater.* **2004**, *16*, 4436.

(54) Wang, A.; Kymissis, I.; Bulovic, V.; Akinwande, I. A. *IEEE Trans. Electron Devices* **2006**, *53*, 9.

(55) Pernstich, K. P.; Haas, S.; Oberhoff, D.; Goldmann, C.; Gundlach, D. J.; Batlogg, B.; Rashid, A. N.; Schitter, G. *J. Appl. Phys.* **2004**, *96*, 6431.

(56) Ryu, R.; Kymissis, I.; Bulovic, V.; Sodini, C. G. *IEEE Electron Device Lett.* **2005**, *26*, 716.

(57) (a) Unni, K. N. N.; Dabos-Seignon, S.; Nunzi, J.-M. *J. Mater. Sci.* **2006**, *41*, 317. (b) Rolland, A.; Richard, J.; Kleider, J.-P.; Mencaraglia, D. *J. Electrochem. Soc.* **1993**, *140*, 3679.

(58) McDowell, M.; Hill, I. G.; McDermott, J. E.; Bernasek, S. L.; Schwartz, J. *Appl. Phys. Lett.* **2006**, *88*, 073505/1.

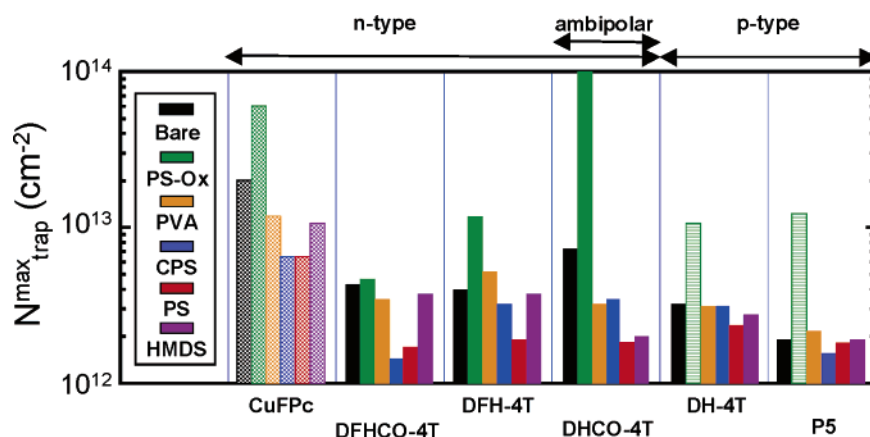


Figure 12. Histogram showing the maximum estimated (according to eq 3) interface trap densities ($N_{\text{max_trap}}^{\text{max}}$) for various semiconductor–dielectric combinations. The dotted plots for CuFPc denote that this semiconductor exhibits substantial I_{off} currents indicating unintentional electron doping. Similarly, the crosshatched lines for DH-4T/PS-Ox and P5/PS-Ox are due to stress at the high I_{off} currents (extra mobile holes) recorded for these devices due to dielectric electron trapping (see text).

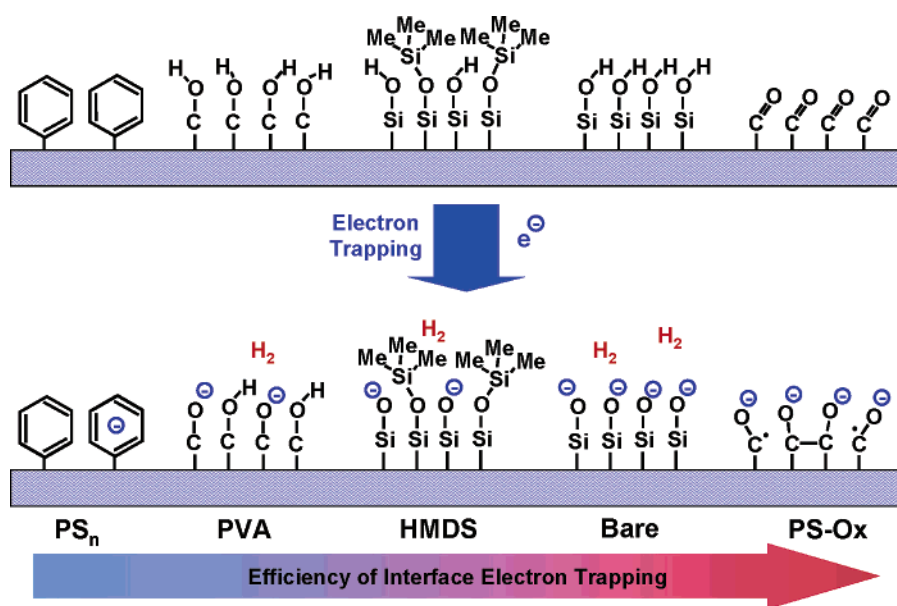


Figure 13. Schematic diagram of functional group electron trapping efficiency on various bilayer dielectric layers.

for the air-sensitive n-type semiconductors and p-type DH-4T in view of the similarities demonstrated in the semiconductor film properties. Therefore, a recognizable correlation might be expected between the mobility enhancement effects and the estimated trap densities, in turn dependent on the dielectric surface modifications. In the following, we discuss, first, OTFT devices based on high-lying LUMO n-type semiconductors which exhibit significant modulation of device performance via dielectric surface modification, followed by analysis of the air-stable n-type and p-type devices.

Interestingly, it can be seen that similar trap density patterns are observed among the air-sensitive n-type semiconductors although trap density comparisons using eq 3 are formally valid only for transistors having identical semiconductors. The apparent trend is that PS-Ox exhibits the greatest trap densities followed by Bare or PVA, while CPS and PS1 exhibit the lowest densities. In addition, the trend in interface trap density as a function of bilayer dielectric structure exhibits a close correlation with TFT mobility for the n-type semiconductors having high-lying LUMOs. Thus, in the case of DFHCO-4T, DFH-4T, and DHCO-4T, PS1 devices with the highest

mobilities exhibit the lowest interface trap densities and PS-Ox devices with the lowest mobilities exhibit the largest interface trap densities.

Although the nature of interface traps is doubtless dependent on intricate microstructural details of the interaction between the semiconducting and dielectric layers, the interface trap density can be qualitatively understood from a chemical perspective (Figure 13). It is known that, in the absence of special surface modifications, Bare substrates exhibit an interface trap density of $\sim 10^{12} \text{ cm}^{-2}$ which is principally attributed to interfacial chemical functionalities/species such as Si–OH, in conjunction with adsorbed H_2O , and adventitious carbon contamination.^{59,60} Such chemical defects can affect charge transport by deep-trapping/doping and/or by scattering carriers at the semiconductor–dielectric interface, and this effect is reflected in device performance parameters. Such interfacial effects are important in organic semiconductor-based devices since the charge transport process is believed to occur within the very first few semiconducting layers in proximity to the

(59) Alexandrova, S.; Sxekeres, A. *Phys. Status Solidi A* **2001**, *187*, 499.

(60) Olthuis, W.; Bergveld, P. *IEEE Trans. Electr. Insul.* **1992**, *27*, 691.

gate dielectric.¹⁵ Note that the chemical functionalities depicted in Figure 13 should preferentially trap electrons (rather than holes) leading to mobility suppression in n-type semiconductors, and that for p-type semiconductors the principal outcome is introduction of extra holes at the semiconductor–dielectric interface for charge balance and a positive V_T shift.³⁹ The importance of controlling interface traps was demonstrated in the realization of ambipolar pentacene TFTs. Thus, Ahles et al. reported that substantial n-type activity in pentacene TFTs is enabled by compensating electron traps at the semiconductor–dielectric interface via introduction of an ultrathin Ca layer between the pentacene and SiO₂ layers.⁶¹ More recently, Chua et al. reported that n-type activity in known p-type semiconductors (e.g., F8T2) is stabilized by passivating surface electron-trapping silanol groups (Si–OH) on the SiO₂ dielectric and suggested that their electron affinity is ~ 3.9 eV.¹⁶ Note that the LUMO energies of air-sensitive n-type semiconductors **DFHCO-4T**, **DFH-4T**, and **DHCO-4T** lie very close to or above the negative of this value, while that of air-stable n-type **CuFPc** is far lower (Figure 11B). Indeed, the mobility enhancement observed on passivating trap-active SiO₂ surfaces using HMDS vs **Bare** is significantly more pronounced for semiconductors with higher LUMO energies (**DFHCO-4T** < **DFH-4T**, **DHCO-4T**), while **CuFPc** exhibits very similar mobilities on **Bare** and **HMDS** (Figure 11A), implying that electron-trapping on surface-modified dielectrics is related to the ambient sensitivity/LUMO energies of n-type semiconductors and to the chemical nature of the dielectric surface. Note that although it was previously suggested that elusive n-type organic semiconductor behavior is governed by interface traps, especially on SiO₂,^{16,40} the present analysis is, to our knowledge, the first study to explicitly relate dielectric surface chemistry to OTFT performance enhancement via the semiconductor FMO energetics and interface traps. In the following, we address the suppressed/enhanced device performance on other dielectric structures from the standpoint of surface chemistry and interface trap energetics.

The very significant interface trap reduction on **PSn** is supported by several mechanistic arguments: (1) effective passivation of surface chemical defects with a very smooth morphology and minimal nanoscopic pinholes, (2) possible stabilization of charge carriers at the dielectric interface by aromatic cores. Regarding improved device stability and performance, Me₃Si-terminated HMDS-treated dielectric surfaces have been shown to generally enhance OTFT performance.²⁴ This strategy has been extended to dielectric surface modification using long-chain organosilanes such as octadecyltrichlorosilane (OTS).²⁵ The origin of the device performance improvement by this organosiloxane interface modification is doubtless the substantial reduction in electron-trapping silanol density on the SiO₂ surface without affecting the dielectric surface morphology/roughness.¹⁶ However, it has been shown that silanol groups cannot be completely removed by such self-assembled alkyl layers⁶² and that the relatively thin self-assembled alkyl monolayers (<2 nm) are subject to charge carrier tunneling.⁶³ Note also that the larger advancing aqueous contact angle on **HMDS** (102°) versus **PSn** (92°) does not

necessarily mean complete surface coverage.⁶⁴ Furthermore, recent X-ray photoelectron spectroscopic studies of HMDS–SiO₂ surface modification⁶⁵ demonstrate that, even after careful HMDS treatment, adventitious carbon contaminants,⁶⁶ including some carbonyl groups, which are efficient electron traps (vide infra), still remain. In contrast, surface modification with electrically “inert” and relative thick PS coatings (>24 nm vs <1 nm for HMDS) should completely cover the SiO₂ surface, contain minimal pinholes, and more effectively cover/passivate “trap-generating” surface functionalities without significant changes in surface morphology ($\rho \approx 0.3$ nm). Note also that the LUMO level of PS, estimated from the electrochemical reduction potential of benzene,⁶⁷ is ~ -1.5 eV, too high to act as an electron trap.

Self-assembled monolayers having phenyl or fused-arene termini on SiO₂ dielectric surfaces are claimed to stabilize charge transport in pentacene OTFTs. Pernstich et al. reported that pentacene TFTs on SiO₂ modified with self-assembled monolayers having pendent phenyl groups exhibit low off currents and low subthreshold voltage swings.⁵⁵ It was reported that pentacene transistors based on phenyltrichlorosilane-modified dielectrics exhibit good hole mobilities (~ 0.7 cm²/(V s)) and low off current levels (10⁻¹² A) at zero gate bias, in contrast to devices on self-assembly modified SiO₂ using phenyl groups with electron-withdrawing, dipolar substituents.⁵⁵ More recently, SiO₂ dielectric surface modification with self-assembled anthracene layers has also been reported to reduce charge trapping state densities as well as subthreshold voltage swings.⁵⁸ The interaction between arene-modified surfaces and organic semiconducting films is not fully understood; however we hypothesize that similar interactions may occur between semiconductor and PS layers, considering the chemical similarities. Although spin-coated PS layers are expected to have more random orientations of phenyl substituents with respect to the surface, this “soft” surface may better conform to semiconductor crystal growth patterns, and the surface coverage should be complete with minimal pinholes, in contrast to self-assembled systems where incomplete coverage or local structural defects may occur.⁶²

As shown in Figures 11 and 12, **PVA**-based devices exhibit lower enhancement factors (η) and higher trap densities than **PS**-based devices. One possible explanation concerns the influence of the dielectric constant. Veres et al. proposed that amorphous semiconducting polymers exhibit higher performance parameters on polymeric insulators with low dielectric constants than on those with high dielectric constants, and argued that local polarization effects in high dielectric constant polymers may induce the localization of carrier states by dipolar disorder.³⁰ Such dipolar disorder effects were also proposed in OTFTs using polycrystalline⁶⁸ and single-crystalline organic semiconductors.⁶⁹ Considering the relatively low dielectric constant of **PS** ($k = 2.5$) and **CPS** layers ($k = 2.6$) vs **PVA** ($k = 7.4$), it is possible

(61) Ahles, M.; Schmechel, R.; von Seggern, H. *Appl. Phys. Lett.* **2004**, *85*, 4499.

(62) Angst, D. L.; Silmmons, G. W. *Langmuir* **1991**, *7*, 2236.

(63) Wasserman, S. R.; Tao, Y. T.; Whitesides, G. M. *Langmuir* **1989**, *5*, 1074.

(64) Shiu, J.-Y.; Kuo, C.-W.; Chen, P.; Mou, C.-Y. *Chem. Mater.* **2004**, *16*, 561.

(65) Zen, A.; Neher, D.; Silmy, K.; Hollander, A.; Asawapirom, U.; Scherf, U. *Jpn. J. Appl. Phys.* **2005**, *44*, 3721.

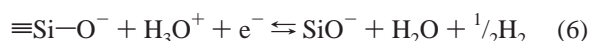
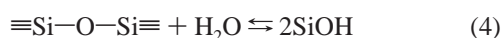
(66) Ogata, T.; Ban, C.; Ueyama, A.; Muranka, S.; Hayashi, T.; Kobayashi, K.; Kobayashi, J.; Kurokawa, H.; Ohno, T.; Hirayama, M. *Jpn. J. Appl. Phys.* **1998**, *47*, 2468.

(67) Mortensen, J.; Heinze, J. *Angew. Chem., Int. Ed. Engl.* **1984**, *23*, 84.

(68) Uni, K. N. N.; Dabos-Seignon, S.; Nunzi, J.-M. *J. Mater. Sci.* **2006**, *41*, 317.

(69) Stassem, A. F.; de Boer, R. W. I.; Iosad, N. N.; Morpurgo, A. F. *Appl. Phys. Lett.* **2004**, *85*, 3899.

that the reduced carrier localization/trapping at the low k dielectric interface contributes to low interface trap densities and enhanced electron mobilities. In contrast, other results suggest that high k dielectric materials increase mobility.^{13g,30b} In any case, note that C–OH groups should be differentiated from Si–OH groups in terms of electron trapping. It has been proposed that the electron trapping mechanism on the SiO₂ surface involves electron capture by protons and release of H₂, suggesting that the stable negative charges are formed on SiO₂ and that traps are consumed (eqs 4–6).⁶⁰ The acidity of the hydroxyl group plays a key role in shifting the equilibrium toward the right, hence enabling electron trapping.



Note that a mechanism based on similar reactions is plausible for alcohols. However, alcohols and phenols (C–OH) are less efficient proton donors ($\text{p}K_{\text{a}} \approx 10\text{--}15$) and less likely to capture electrons compared to silanols ($\text{p}K_{\text{a}} \approx 5$). In fact, as shown in Figure 11, all n-type semiconductors with high-lying LUMO levels exhibit higher mobilities on **PVA** than on **Bare** by a factor of 2–8, arguing that the presence of alcohol (not silanol) functionalities is less detrimental to n-type charge transport and that the enhanced n-type mobility may be due to the passivation of surface silanol groups as in the case of **PSn**. Note that n-type transport was previously demonstrated in pentacene^{17a} and polymer-blend-OTFTs^{11e} on PVA dielectrics.

Considering that the semiconducting film morphologies and microstructures on **PS-Ox** are very similar to those on the other bilayer dielectric structures, the very poor OTFT performance observed on **PS-Ox**, even compared to that on **Bare**, must be related to the semiconductor–dielectric interface. Despite short exposure times with minimal power (5 s, 20 W), plasma-induced reactions are expected to generate oxidation-related surface chemical defects, although the gross **PS-Ox** surface morphology remains smooth ($\rho \approx 0.3$ nm) and the capacitance is almost identical to that of **PS1** (Table 1).⁷⁰ As shown in Figure 12, a relatively high trap density is estimated on this substrate for all semiconductor combinations ($>10^{13}$ cm²) and can be ascribed to the chemical functionalities generated by the O₂ plasma treatment.⁵⁴ Previous studies of O₂ plasma-treated PS surfaces using ATR-FTIR and NEXAFS found that the density of oxidized-carbon functionalities, especially carbonyl groups, is significant.³⁵ Carbonyl groups should have substantial electron affinities, and based on the electrochemical reduction potentials⁷¹ of acetophenone (−1.99 V), benzophenone (−1.72 V), and methylvinylketone (−1.11 V) versus SCE, the estimated LUMO energies of carbonyl functionalities generated by the O₂ plasma should lie within −2.8 to −3.7 eV. This range of LUMO energies is very close to those of the air-sensitive n-type semiconductors, and carbonyl groups on the dielectric surface are therefore expected to strongly perturb electron transport at the semiconductor–dielectric interface by trapping electrons. This effect is most pronounced in the case of **DHCO-4T** on

PS-Ox where n-type activity completely disappears. Note also that the estimated trap density on **PS-Ox** is much greater than that on **Bare**, indicating that the carbonyl groups more efficiently trap electrons than silanol. In the case of p-type semiconductors, the effect of electron trapping on device performance is primarily the positive shift of V_{T} , and its extent is far more pronounced on **PS-Ox** than on **Bare** as shown in Figure 5 (vide infra).

In addition to mobility modulation, the current–voltage hysteresis evident in the present transfer plots can be tuned by employing different bilayer dielectrics. Thus, while PVA films are known to act as insulators and exhibit hysteresis which presumably arises from charge storage and polarization, PVA efficacy as a polymeric gate electret has also been demonstrated.⁴⁰ In contrast, the more remarkable hysteretic behavior on **PS-Ox** substrates (Figure 5) must be related to a very large density of deep interface traps. In the case of n-type semiconductors on **PS-Ox**, the interface trap states are estimated from the subthreshold swing to be $\sim 10^{13}$ cm^{−2}, much greater than those on the other bilayer structures and cause serious current–voltage hystereses by binding electrons in these rather deep traps at the semiconductor–dielectric interface. Also, this type of electron trapping is partially indicated by gate-bias dependent semiconductor mobilities in the n-type TFTs on **PS-Ox**. As argued from the AFM morphology results on the very thin films of these semiconductors, the observed mobility variations are largely governed by dielectric surface chemistry rather than by gross film morphology or a growth mechanism, and such semiconductor–dielectric interactions are clearly revealed by the interface trap densities.

From the discussion of surface-chemistry-related interface trap energetics, it is arguable that the relative insensitivity of air-stable n-type **CuFPc** performance to the dielectric surface chemistry can be ascribed to the very low-lying LUMO. In contrast to the air-sensitive n-type semiconductor devices, air-stable **CuFPc** devices do not exhibit direct correlations between the observed semiconductor mobilities and the (overall high) estimated trap density ($>10^{13}$ cm^{−2}), although the trap density variation pattern is similar to those of the air-sensitive n-type semiconductors. The reason is likely an overestimated trap density due to charge trapping in the semiconducting layer by chemical impurities. Note that **CuFPc** exhibits a very low-lying LUMO energy, also indicating that this semiconductor is vulnerable to reversible/irreversible doping by chemical impurities rather than to the aforementioned interface chemical functionalities. Indeed, **CuFPc** TFTs exhibit relatively high I_{DS} currents at zero gate bias (Figure 4), and the theoretically estimated charge accumulation density at the semiconductor–dielectric interface at $V_{\text{G}} = 50$ V (3×10^{12} cm^{−2}, $n = Q/e = (C_{\text{i}} \times V_{\text{G}})/e$) is far lower than the estimated trapped charge density in Figure 12. Therefore, the estimated trap density includes traps both in the semiconducting layer and at the semiconductor–dielectric interface. Considering that **CuFPc** films on different dielectrics exhibit different morphologies, the observed mobility modulation in the various bilayer dielectric structures can be attributed to the combined effects of bulk semiconductor film morphology/doping and the semiconductor–dielectric interface.

The estimated trap densities of the present p-type semiconductors, except on the **PS-Ox** substrates, are almost constant, which is opposite to the air-sensitive n-type semiconductor

(70) Teyssedre, G.; Laurent, C. *IEEE Trans. Dielectr. Electr. Insul.* **2005**, *12*, 5.

(71) Fry, A. J. *Synthetic Organic Electrochemistry*; Harper & Row: New York, 1972.

trends (Figure 12). Since **DH-4T** exhibits essentially the same film morphology and microstructure on all of the present bilayer dielectrics, the similarity in calculated trap densities can be ascribed to invariant semiconductor–dielectric interfacial properties. Therefore, the nearly constant hole mobilities of **DH-4T** on the different dielectrics can be correlated with the relative insensitivity of hole transport to the dielectric surface nature. Note that in the case of **DH-4T** on **PS-Ox**, which exhibits much greater interface trap density than those on other dielectric surfaces, V_{on} and V_{T} are shifted positively compared to that on **PS1** and **PVA**. The off current is also significantly increased due to the extra holes created by electrons filling the interface traps, and the relatively large hysteresis can be explained in a similar way.³⁹ In contrast, the estimated trap densities in pentacene devices do not offer a clear picture of interface trapping. Although similar trends in estimated trap densities are apparent in pentacene devices on the bilayer dielectric structures, differences in film morphologies and microstructures revealed by SEM and XRD indicate that the estimated trap densities are affected by the grain boundaries in the semiconducting layer, molecular orientation, degrees of texture, and the surface chemistry of the semiconductor–dielectric interface. For example, although pentacene TFTs on **PVA** exhibit relatively low trap densities, which are very similar to those on other bilayer structures except **PS-Ox**, OTFT performance is poorest even compared to that on **PS-Ox** which exhibits very high trap densities. As demonstrated by the SEM and advancing aqueous contact angle data, such inferior device performance can be principally attributed to small, discontinuous grains in the semiconducting film. Therefore, unlike the air-sensitive n-type semiconductors and **DH-4T**, which exhibit relatively similar solid-state film behavior on different bilayer dielectrics, assessment of trap densities here involves changes in semiconductor properties such as grain sizes as well as in the semiconductor–dielectric interface.

For the semiconductor–dielectric combinations described here, we are currently performing admittance (ac) measurements in metal–insulator–(organic)semiconductor capacitors over a large range of applied bias and frequency to achieve a quantitative distribution of the interface trapping states as a function of energy.⁷²

Conclusions

The present contribution describes a general approach to probe OTFT semiconductor–dielectric interface chemical effects on transistor performance parameters using tailored bilayer dielectrics. Very different organic semiconductors with p-/n-type and ambipolar charge transport characteristics are grown on six different bilayer dielectric structures and systematically characterized by AFM, SEM, advancing aqueous contact angles, and WAXRD. In concert, the corresponding transistor device response parameters are investigated in detail. Polystyrene coatings on SiO_2 with minimal gate leakage and surface roughness, significantly enhance the mobilities of air-sensitive n-type semiconductors, while such kinds of device performance improvement is nominal in the case of air-stable n-type and p-type semiconductors. Based on interface trap density estimations on the various bilayer structures, electron trapping at the semiconductor–dielectric interface is identified as the origin of the mobility sensitivity to the different surface chemistries in the n-type semiconductor systems having high-lying LUMOs. The present semiconductors generally exhibit very similar film morphologies and microstructures, regardless of the dielectric surface modification. This result demonstrates that controlling dielectric interface chemistry is critical in achieving good n-type performance in OTFT configurations and that the proper choice of n-type semiconductor can be utilized to probe interface properties such as trap types and densities on various dielectric surfaces, without perturbation of the semiconducting film morphology and microstructures on top.

Acknowledgment. We thank AFOSR (STTR FA 9550-04-0080), the NSF-MRSEC program through the Northwestern Materials Research Center (DMR-0076097) and the NASA Institute for Nanoelectronics and Computing (NCC 2-3163) for support of this research.

Supporting Information Available: Figures S1–S6. This material is available free of charge via the Internet at <http://pubs.acs.org>.

JA063290D

- (72) (a) Torres, I.; Taylor, D. M.; Itoh, E. *Appl. Phys. Lett.* **2004**, *85*, 314. (b) Saks, N. S.; Ancona, M. G.; Chen, W. *Appl. Phys. Lett.* **1992**, *60*, 2261. (c) Nicollian, E.; Goetzberger, A. *Bell Syst. Tech. J.* **1967**, *46*, 1055.